Lecture Questions: Introduction to Water Pollution

- 1. Where is the water? List the major water reservoirs on Earth (the *hydrosphere*), in order of volume (largest to smallest).
- 2. What is the *hydrologic cycle*?
- 3. Define the following terms:
 - a) solution, solute, and solvent
 - b) concentration (and include common units)
 - c) ionic compound, cation and anion
 - d) electrolyte and nonelectrolyte
- 4. What happens when an ionic compound—like sodium chloride (table salt)—dissolves? How is it different when a molecular compounds—like glucose (sugar)—dissolves?
- 5. *External research.* What are the seven most concentrated ions—four cations, three anions—in natural surface water bodies?
- 6. Define the following acronyms, and describe their significance for drinking water safety:
 - a) MCLG
 - b) MCL
 - c) THMs
- 7. a) List the steps used in treating drinking water; briefly describe the purpose of each step.
 - b) List three disinfection methods, and compare their advantages and disadvantages.
- 8. What is 'water hardness?'