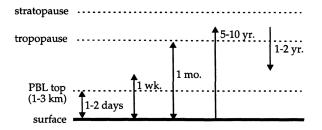
# The Atmosphere

Composition and Structure



# **Outline of Topics**

### Intro to Air Pollution

### 2 Atmospheric Composition

- Concentration of Gases
- Particulate Matter

### 3 Structure of the Atmosphere

- Thermal Stratification
- Mixing Times
- Spatial Variability of Composition

### 🕨 Light

- Nature of Light
- Sunlight

What are the major problems due to pollutants discharged into the atmosphere?

- Stratospheric ozone depletion (due to CFCs, HCFCs, etc)
- Global climate change (due to GHGs, etc)
- Acid deposition (SO<sub>2</sub>, NO<sub>x</sub>)
- Smog (VOCs, NO<sub>x</sub>)
- Particulates (PM, especially 'fine PM')
- Other toxic air pollutants (eg, CO, Pb, Hg, PAHs and other toxic organics, etc)

### Lecture Question

What are the *criteria pollutants*? And what the heck IS a 'criteria pollutant', anyway?

- Carbon monoxide, CO
- Nitrogen dioxide, NO<sub>2</sub>
- Ozone, O<sub>3</sub>
- Lead, Pb
- Particulates, PM<sub>10</sub> and PM<sub>2.5</sub>
- Sulfur dioxide, SO<sub>2</sub>

# Atmospheric Composition

#### Lecture Question

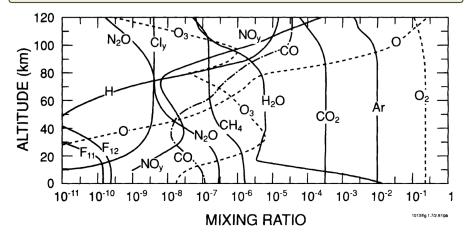
What is in "air?" What is it made of?

The five main gases, with their concentrations, are:

- 1. Nitrogen, N<sub>2</sub>: 78.080%
- 2. Oxygen, O<sub>2</sub>: 20.95%
- 3. Water vapor,  $\rm H_2O:$   $\sim 2.5\%$  (variable, up to 4%)
- 4. Argon, Ar: 0.93%
- 5. Carbon dioxide,  $CO_2$ : 0.04% (398.78 ppm 12/2014 avg at NOAA Mauna Loa observatory)

## Vertical Concentration Profiles

Does the composition of air stay the same as the altitude changes?

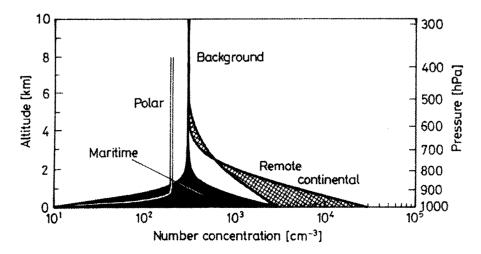


What a minute: the concentration of  $\rm O_2$  in air looks constant ( $\sim 20\%$ ) to an altitude of about 80 km. So why is it hard to breathe on top of Mt. Everest (7.5 km altitude)?

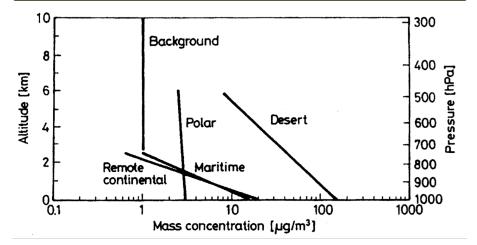
altitude, km	pressure, atm	integrated mass %
0	1.000	0.00 %
10	0.287	71.35 %
15	0.153	84.66 %
25	0.044	95.61%
50	0.002	99.81 %
100	0.000	100.00 %

Note: diameter of the earth is 6400 km.

We've talked about gases; how much *particulate matter* (PM) is there in the atmosphere? And why should we care?



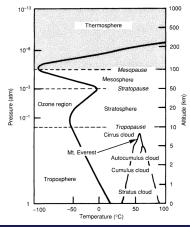
We've talked about gases; how much *particulate matter* (PM) is there in the atmosphere? And why should we care?



# Structure of the Atmosphere

#### Lecture Question

What are the main regions (layers) of the atmosphere? Why do these layers form?

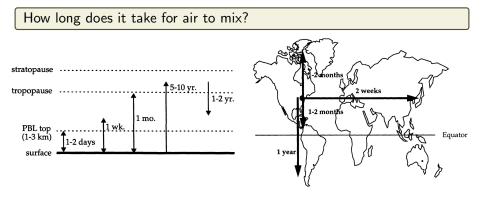


• Troposphere: 0–15 km

Lower 1–3 km is the *planetary boundary layer* (PBL.)

- Stratosphere: 15–50 km
- Mesosphere: 50-100 km
- Thermosphere: 100+ km

## Time Scales of Atmospheric Mixing



### So why are mixing rates important?

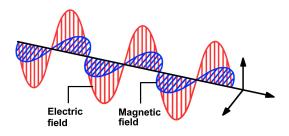
- An important factor in determining the *spatial variability* of atmospheric composition.
- Atmospheric removal rate of a given substance are mainly determined by three properties.

Chemicals that do not absorb sunlight, are not reactive, and are not soluble will tend to have long atmospheric lifetimes.

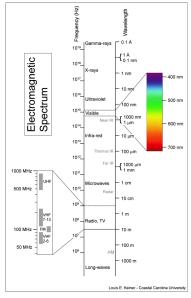
• Mixing *vs* removal determines spatial scale of the effects of a given pollutant.

### Lecture Question

How does light act like a wave? What is the mathematical relationship between wavelength, frequency, and speed of propagation?



# The Electromagnetic Spectrum



#### Lecture Question

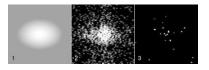
What wavelengths correspond to the ultraviolet and visible regions of the EM spectrum?

Wavelengths	EM Region	
$< 10{ m pm}$	gamma ray	
10 pm – 10 nm	x-ray ("hard" and "soft")	
10 – 200 nm	vacuum ultraviolet	
200 – 380 nm	near ultraviolet	
380 – 750 nm	visible	
$0.75-1.4\mu{ m m}$	near infrared	
$1.4-3\mu{ m m}$	short-wave infrared	
$3 - 8 \mu \text{m}$ mid-wave infrared		
$8 - 15 \mu \text{m}$ long-wave infrared		
$15-1000\mu{ m m}$	far infrared	
1 mm – 1 cm microwave		
1 cm – 10 m	– 10 m radiofrequency	

# Light As a Particle

#### Lecture Question

How does light behave as a particle? What is the relationship between energy and wavelength?



images of decreasing light intensity

• Light imparts energy in quantized amounts

$$E = h\nu = rac{hc}{\lambda}$$

• Photon energy *increases* as frequency *increases* and wavelengths get *shorter* 

# Incident Sunlight

#### Lecture Question

What fraction of sunlight is in the UV, visible, and infrared regions of the spectrum?

