Physics 132-2 Test 2

I pledge that I have neither given nor received unauthorized assistance during the completion of this work.

Name.

Signature.

Questions (5 for 8 pts. apiece) Answer in complete, well-written sentences WITHIN the spaces provided.

1. Use Newton's law of universal gravitation to show that the magnitude of the acceleration due to gravity on an object of mass m at a height h above the surface of the earth is given by the following expression

$$\frac{GM_e}{(R+h)^2}$$

where G is the gravitational constant, M_e is the mass of the Earth, and R is the radius of the Earth. Hint: Because of the spherical symmetry of the Earth you can treat the mass of the Earth as if it were all concentrated at a point at the Earth's center. Explain.

2. For non-uniform forces (that is, forces that change with position), the work done by a force is given by

$$W = -\int \vec{F} \cdot d\vec{s}$$

which reduces to $W = \int F ds$ in one dimension. The diagram below shows the resulting general relationship between force \vec{F} and potential energy U. Complete the equation on the right showing how to find the force if you know the potential energy U. Explain you reasoning.



DO NOT WRITE BELOW THIS LINE.

3. Consider the circuit shown below. How would you connect an ammeter to measure the total current in the circuit? A sketch might be helpful. Explain.



4. In calculating the magnetic field due to a nearby line of current of length L like the one shown in the figure, you encounter the integral shown below

$$|\vec{B}| = \oint \frac{\mu_o \mathcal{I}}{4\pi} \frac{y}{(x^2 + y^2)^{(3/2)}} dx$$

where \mathcal{I} is the current and the point P is over the midpoint of the line of current. What is the direction of the field at point P? Explain.



5. Consider the following idea for a novel propulsion for a ship or submarine. In this 'magnetohydrodynamic drive' seawater flows between the poles of a magnet as shown below and an electric field drives a current through the seawater. The magnetic force on this current propels the water towards the rear of the ship pushing the ship forward. What should be the direction of the applied electric field in the region between the poles of the magnet? Explain.



Problems (3). Clearly show all reasoning for full credit. Use a separate sheet to show your work.

1. 15 pts. The x axis is the symmetry axis of a stationary, uniformly charged ring of radius R and charge Q (see figure). A particle with charge Q and mass M is located on this axis. Starting from the electric potential of a point charge, what is the electric potential along the x-axis in terms of x, Q, R, M, and any other constants?



Problems (3). Clearly show all reasoning for full credit. Use a separate sheet to show your work.

2. 20 pts. Many mass spectrometers have an additional device to insure the velocity of the particles entering the magnetic field of the spectrometer is the desired value. This 'velocity selector' consists of an electric field and a second magnetic field oriented perpendicular to the \vec{E} field (see figure below). These two fields are set so that only particles with the desired velocity are undeflected as they move through the device. Negative ¹⁴C ions with charge q = -e are accelerated across the electric potential $V = 10^6 V$ and enter the velocity selector. The magnetic field of the velocity selector is 0.5 T. To pick out the ¹⁴C ions, what should be the value of the electric field in the velocity selector and how should it be oriented?



3. 25 pts. In a fusion experiment a magnetic 'bottle' holds a gas of hydrogen atoms with an initial density $\rho = 2.8 \times 10^{21} atoms/m^3$. After reaching a high temperature the hydrogen gas within this volume becomes partially ionized meaning that each proton spends part of its time bound to an electron and part of the time alone. An applied electric field causes the bare protons to drift in a cylindrical region of cross sectional area $A = 1.2 \times 10^{-3} m^2$ at $v_p = 25 m/s$ while the electrons drift in the opposite direction at $v_e = 88 m/s$ (see figure). The net current has magnitude $|\mathcal{I}| = 21 A$. What fraction of the gas is ionized?



DO NOT WRITE BELOW THIS LINE.

Physics 132-02 Equations Test 2

$$\vec{F}_{G} = -G \frac{m_{1}m_{2}}{r_{12}^{2}} \hat{r} \qquad \vec{F}_{C} = k_{e} \frac{q_{1}q_{2}}{r_{12}^{2}} \hat{r} \qquad \vec{E} \equiv \frac{\vec{F}}{q_{0}} \qquad \vec{E} = k_{e} \sum_{i} \frac{q_{i}}{r_{i}^{2}} \hat{r}_{i} \qquad \vec{E} = k_{e} \int \frac{dq}{r^{2}} \hat{r} \qquad k_{e} = \frac{1}{4\pi\epsilon_{0}}$$

$$\vec{E}_{dipole} = k_{e} \frac{q(2a)}{(x^{2} + a^{2})^{3/2}} \hat{j} \qquad \vec{E}_{ring} = k_{e} \frac{qx}{(x^{2} + R^{2})^{3/2}} \hat{i} \qquad \vec{E}_{plane} = 2\pi k_{e} \eta \hat{k} = \frac{\eta}{2\epsilon_{0}} \hat{k}$$

$$\vec{E}_{disk} = 2\pi k_{e} \eta \left[1 - \frac{z}{\sqrt{z^{2} + R^{2}}} \right] \hat{k} = \frac{\eta}{2\epsilon_{0}} \left[1 - \frac{z}{\sqrt{z^{2} + R^{2}}} \right] \hat{k}$$

$$W \equiv \int \vec{F} \cdot d\vec{s} \quad \Delta V \equiv \frac{\Delta PE}{q_{0}} = -\int_{A}^{B} \vec{E} \cdot d\vec{s} \quad V = k_{e} \frac{q}{r} \quad V = k_{e} \sum_{i} \frac{q_{i}}{r_{i}}$$

$$V = k_e \int \frac{dq}{r} \quad V = Ed \quad I = \frac{dQ}{dt} \quad Q = \int Idt \quad V = IR \quad P = IV \quad R_{equiv} = \sum R_i$$

The algebraic sum of the potential changes
$$I = nqv_d A$$

across all the elements of a closed loop is zero.

$$\begin{split} \vec{F}_{B} &= q\vec{v} \times \vec{B} \quad |\vec{F}_{B}| = |qvB\sin\alpha| \quad \vec{B} = k_{m} \int \frac{Id\vec{s} \times \hat{r}}{r^{2}} \quad k_{m} = \frac{\mu_{0}}{4\pi} \quad \vec{B}_{ring} = \frac{\mu_{0}IR^{2}}{2} \frac{1}{(x^{2} + R^{2})^{3/2}} \hat{i} \\ KE_{0} + PE_{0} = KE_{1} + PE_{1} \quad KE = \frac{1}{2}mv^{2} \quad PE = qV \\ \vec{F} &= m\vec{a} \quad |\vec{F}_{cent}| = m\frac{v^{2}}{r} \qquad x = \frac{a}{2}t^{2} + v_{0}t + x_{0} \qquad v = at + v_{0} \\ \frac{dx^{n}}{dx} = nx^{n-1} \qquad \frac{df(u)}{dx} = \frac{df}{du}\frac{du}{dx} \qquad \frac{d}{dx}f(x) \cdot g(x) = f\frac{dg}{dx} + g\frac{df}{dx} \\ \langle x \rangle &= \frac{1}{N}\sum_{i}x_{i} \quad \sigma = \sqrt{\frac{\sum_{i}(x_{i} - \langle x \rangle)^{2}}{N-1}} \quad A = 4\pi r^{2} \quad V = Ah \quad V = \frac{4}{3}\pi r^{3} \\ \vec{A} \cdot \vec{B} = A_{x}B_{x} + A_{y}B_{y} + A_{z}B_{z} \quad \vec{A} \times \vec{B} = |\vec{A}||\vec{B}||\sin\alpha| \rightarrow rhr \\ \ln(ab) &= \ln a + \ln b \quad \ln\left(\frac{a}{b}\right) = \ln a - \ln b \quad \ln x^{n} = n\ln x \quad x = e^{\ln x} = \ln(e^{x}) \\ \frac{df(x)}{dx} &= \lim_{\Delta x \to 0} \frac{f(x + \Delta x) - f(x)}{\Delta x} \quad \int_{a}^{b} f(x)dx = \lim_{\Delta x \to 0} \sum_{n=1}^{N} f(x)\Delta x \quad \frac{d\ln x}{dx} = \frac{1}{x} \\ \int \frac{1}{x}dx = \ln x \quad \int x^{n}dx = \frac{x^{n+1}}{n+1} \quad \int e^{ax}dx = \frac{e^{ax}}{a} \quad \int \frac{1}{\sqrt{x^{2} + a^{2}}}dx = \ln\left[x + \sqrt{x^{2} + a^{2}}\right] \end{split}$$

$$\int \frac{x}{\sqrt{x^2 + a^2}} dx = \sqrt{x^2 + a^2} \qquad \int \frac{x^2}{\sqrt{x^2 + a^2}} dx = \frac{1}{2}x\sqrt{x^2 + a^2} - \frac{1}{2}a^2 \ln\left[x + \sqrt{x^2 + a^2}\right]$$
$$\int \frac{x^3}{\sqrt{x^2 + a^2}} dx = \frac{1}{3}(-2a^2 + x^2)\sqrt{x^2 + a^2} \qquad \int \frac{1}{(x^2 + a^2)^{3/2}} dx = \frac{x}{a^2\sqrt{x^2 + a^2}}$$
$$\int \frac{x}{(x^2 + a^2)^{3/2}} dx = \frac{-1}{\sqrt{x^2 + a^2}} \qquad \int \frac{1}{x^2 + a^2} dx = \frac{1}{a}\tan^{-1}\left(\frac{x}{a}\right)$$

Physics 132-3 Constants

$T_{boiling}$ (N ₂)	77 K	$T_{freezing}$ (N ₂)	63 K
$T_{boiling}$ (water)	373 K or 100°C	$T_{freezing}$ (water)	273 K or $0^{\circ}\mathrm{C}$
$L_v(\text{water})$	$2.26 imes 10^6 \ J/kg$	L_f (water)	$3.33 imes 10^5 \ J/kg$
$L_v(N_2)$	$2.01 \times 10^5 \ J/kg$	c (copper)	$3.87 \times 10^2 J/kg - ^{\circ} C$
c (water)	$4.19\times 10^3 \ J/kg-K$	$c \; (\text{steam})$	0.69 J/kg - K
c (iron)	$4.5 \times 10^2 J/kg - k$	c (aluminum)	$9.0 \times 10^2 J/kg - K$
ρ (water)	$1.0 \times 10^3 kg/m^3$	P_{atm}	$1.01\times 10^5~N/m^2$
R	8.31J/K - mole	g	$9.8 m/s^2$
0 K	-273° C	Speed of light (c)	$3.0 imes 10^8 \ m/s$
proton/neutron mass	$1.67 \times 10^{-27} \ kg$	k_B	$1.38\times 10^{-23}~J/K$
Gravitation constant	$6.67 \times 10^{-11} N - m^2/kg^2$	1.0 eV	$1.6\times 10^{-19}~J$
e electronic charge	$1.6 \times 10^{-19} C$	Electron mass	$9.11\times 10^{-31}~kg$
Permittivity constant (ϵ_0)	$8.85 \times 10^{-12} \frac{kg^2}{N-m^2}$	$1 \ u$	$1.67\times 10^{-27}~kg$
Permeability constant (μ_0)	$4\pi \times 10^{-7} \ Tm/A$	Earth-Sun distance	$1.5\times 10^{11}~m$
$k_e = 1/4\pi\epsilon_0$	$8.99 \times 10^9 \ N - m^2/C^2$	Earth's mass	$5.97 \times 10^{24} \ kg$
$k_m = \mu_o/4\pi$	$10^{-7} \ Tm/A$	Earth's radius	$6.37 \times 10^6 m$

bydrogon	2 - 2 - T		255		1.5	C	8573		0.57	1.1	6.00	1000	676720	100	0.0276	100	55	holium
1 1																		2
LΩ																		Lla
																		не
1.0079																		4.0026
lithium	beryllium												boron	carbon	nitrogen	oxygen	fluorine	neon 10
3	4												5	6		8	9	10
Li	Be												В	С	N	0	F	Ne
6.941	9.0122												10.811	12.011	14.007	15.999	18.998	20.180
sodium	magnesium												aluminium	silicon	phosphorus	sulfur	chlorine	argon
11	12												13	14	15	16	1/	18
Na	Mg												AI	Si	Ρ	S	CI	Ar
22.990	24.305												26.982	28.086	30.974	32.065	35.453	39.948
potassium	calcium		scandium	titanium	vanadium	chromium	manganese	iron	cobalt	nickel	copper	zinc	gallium	germanium	arsenic	selenium	bromine	krypton
19	20		21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
39.098	40.078		44.956	47.867	50.942	51.996	54.938	55.845	58.933	58.693	63.546	65.39	69,723	72.61	74.922	78.96	79.904	83.80
rubidium	strontium		yttrium	zirconium	niobium	molybdenum	technetium	ruthenium	rhodium	palladium	silver	cadmium	indium	tin	antimony	tellurium	iodine	xenon
37	38		39	40	41	42	43	44	45	46	4/	48	49	50	51	52	53	54
Rb	Sr		Y	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Те		Xe
85.468	87.62		88.906	91.224	92.906	95.94	[98]	101.07	102.91	106.42	107.87	112.41	114.82	118.71	121.76	127.60	126.90	131.29
caesium	barium	F7 70	lutetium	hafnium	tantalum	tungsten	rhenium	osmium	iridium	platinum	gold	mercury	thallium	lead	bismuth	polonium	astatine	radon
55	50	57-70	- 11	12	13	74	15	76		78	79	80	81	82	83	84	85	86
Cs	Ba	*	Lu	Hf	Та	W	Re	Os	lr	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn
132.91	137.33		174.97	178.49	180.95	183.84	186.21	190.23	192.22	195.08	196.97	200.59	204.38	207.2	208.98	209	[210]	[222]
francium	radium	00 400	lawrencium	rutherfordium	dubnium	seaborgium	bohrium	hassium	meitnerium	ununnilium	unununium	ununbium		ununquadium				
87	88	89-102	103	104	105	106	107	108	109	110	111	112		114				
Fr	Ra	* *	Lr	Rf	Db	Sa	Bh	Hs	Mt	Uun	Uuu	Uub		Uua				
[223]	[226]		[262]	[261]	[262]	[266]	[264]	[269]	[268]	[271]	[272]	[277]		[289]				
														<u> </u>				

*Lanthanida sorios	lanthanum 57	cerium 58	praseodymium 59	neodymium 60	promethium 61	samarium 62	europium 63	gadolinium 64	terblum 65	dysprosium 66	holmium 67	erbium 68	thulium 69	ytterbium 70
Lanthannue series	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
	138.91	140.12	140.91	144.24	[145]	150.36	151.96	157.25	158.93	162.50	164.93	167.26	168.93	173.04
* * Actinide series	actinium	thorium	protactinium	uranium	neptunium	plutonium	americium	curium	berkelium	californium	einsteinium	fermium	mendelevium	nobelium
	89	90	91	92	93	94	95	96	97	98	99	100	101	102
	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
	[227]	232.04	231.04	238.03	[237]	[244]	[243]	[247]	[247]	[251]	[252]	[257]	[258]	[259]

The Periodic Chart.