## Physics 132-04 Final Exam

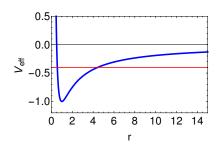
I pledge that I have neither given nor received unauthorized assistance during the completion of this work.

Signature \_\_\_\_\_

Questions (10 for 4 pts. apiece) Answer in complete, well-written sentences WITHIN the spaces provided.

1. Recall the laboratory where you solved the Schroedinger equation. What requirement or postulate forces us to choose particular energy states? Why?

2. Consider the plot below of the effective potential for a macroscopic particle (blue curve) and its line of constant energy (red curve). What happens at the intersection of the energy 'curve' and the effective potential curve?



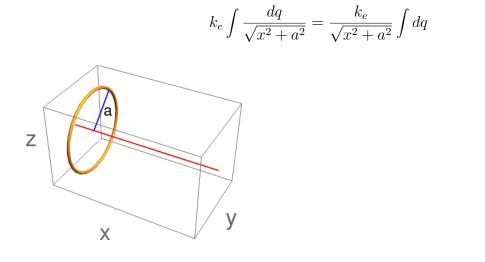
3. Recall the laboratory where you started with a mixture of ice and water in a beaker and heated it it to boiling. What is the relationship between the temperature and the added heat while the ice is melting?

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4. Consider an ideal gas in a rigid container that has a fixed volume. How is the molar specific heat defined in terms of the heat added Q and the temperature change  $\Delta T$ ? If the gas is heated by an amount Q, then how much work is done against the fixed container? Explain.

5. Consider two Einstein solids A and B that are in thermal contact and in an initial macrostate defined by  $N_A$ ,  $q_A$ ,  $N_B$ , and  $q_B$  where N refers to the number of atoms and q refers to the number of quanta in each solid. The initial, total multiplicity is  $\Omega_1$ . One quanta is moved from solid B to solid A and the new, total multiplicity is  $\Omega_2 < \Omega_1$ . Is it more likely for this change to occur or for the system to remain in its initial state? Explain.

6. Recall the calculation of the electric potential due to a charged ring of radius a on the axis x of the ring as shown in the figure. Is the following step legal? Why or why not?



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7. In the mass spectrometer the electron 'falls' across a potential difference created by the accelerating voltage to gain a velocity v before it enters the magnetic field region. Assuming the electron starts from rest, what is the relationship between the accelerating voltage V and the kinetic energy K when it leaves the accelerating region and enters the magnetic field? Justify your answer in words or with equations.

8. Is light a particle or a wave? What is your evidence?

9. Consider the anomalous magnetic moment of the electron. The measured value is

$$a_e = 0.00115965218073(28)$$

where the number in parentheses is the uncertainty on the last two digits. The theoretical value (calculated using quantum mechanics) is

$$a_t = 0.00115965218113(86)$$

where again the number in parentheses is the uncertainty on the last two digits. Do experiment and theory agree? Be quantitative in your answer.

10. Two rooms A and B of equal volume are connected by an open passageway and are maintained at two, different temperatures  $T_A$  and  $T_B$  and  $T_A > T_B$ . Which room has more air molecules? Explain.

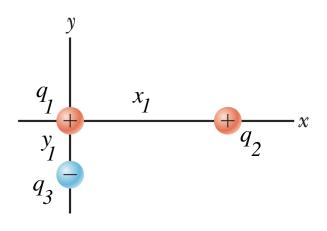
Problems (6). Clearly show all reasoning for full credit. Use a separate sheet to show your work.

- 1. 8 pts. A room of volume  $V_0$  contains air having equivalent molar mass  $M_m$  (in g/mole). If the temperature of the room is raised from  $T_1$  to  $T_2$ , what mass of air  $m_a$  will leave the room? Assume the air pressure is always  $P_0$ .
- 2. 8 pts. A newly-created material has a multiplicity

$$\Omega = \alpha N E$$

where N is the number of atoms in the solid, E is the total, internal energy in the solid, and  $\alpha$  is a constant of proportionality.

- 1. How does the temperature of the new material depend on the internal energy?
- 2. What is the molar heat capacity for this solid?
- 3. Could this material really exist? Why or why not?
- 3. 8 pts. Three point charges are arranged as shown in the figure. What is the vector electric field that  $q_2$  and  $q_3$  create together at the origin? What is the force on  $q_1$ ? Use the  $\hat{i}, \hat{j}, \hat{k}$  notation to express your vectors.



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4. 12 pts. A photon with a wavelength  $\lambda$  is absorbed by an electron (mass  $m_e$ ) confined to a box with infinitely high walls. As a result, the electron moves from the n = 1 state to the n = 4 state. What is the length of the box in terms of  $\lambda$  and any other constants? What is the chance the electron in the n = 4 final state will be found in the left-hand half of the box? The wave function and energies of the electron in the box are

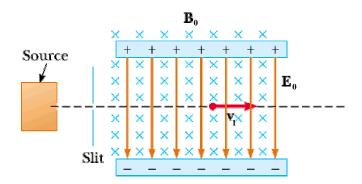
$$\psi(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right) \quad 0 < x < a$$
$$= 0 \qquad x \le 0 \text{ and } x \ge L$$
$$E_n = \frac{n^2 \pi^2 \hbar^2}{2m_e L^2}$$

5. 12 pts. A radar for tracking aircraft must deliver an average intensity of  $I_1 = 6 J/s - m^2$  on a distant aircraft to detect it. It operates at a frequency  $f = 1.2 \times 10^{10} s^{-1}$  from circular radar antenna of radius  $r_a = 1 m$  and emits a power  $P_0 = 10^5 J/s$ . From a wave perspective, the antenna acts as a circular aperture. What is the range of the radar system?



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6. 12 pts. In some magnetic spectrometers a device called a velocity selector is used so the beam particles are all moving at the same velocity when they enter the magnetic field. It consists of crossed electric and magnetic fields as shown in the figure where the electric field is  $\vec{E} = -E_0\hat{j}$  and the magnetic field goes into the plane of the paper  $\vec{B} = -B_0\hat{k}$ . Consider a particle of charge  $q_1$  and mass  $m_1$  that goes through the acceleration phase of the mass spectrometer with a voltage drop  $V_0$ . What is the kinetic energy  $K_1$  of the particle when it emerges in terms of  $V_0$  and  $q_1$ ? For a magnetic field of magnitude  $B_0$  what is value of the electric field magnitude  $E_0$  where the charged particle will pass through the velocity selector without a change in direction (*i.e.* straight through)? Will a second particle that goes through the acceleration phase with the same charge  $q_1$ , but a different mass  $m_2$  also go through undeviated?



Physics 132-4 Constants

Avogadro's number $(N_A)$	$6.022\times 10^{23}$	Speed of light $(c)$	$3 \times 10^8 \ m/s$
$k_B$	$1.38 \times 10^{-23} \ J/K$	proton/neutron mass	$1.67\times 10^{-27}~kg$
1 u	$1.67\times 10^{-27}~kg$	g	$9.8 \ m/s^2$
Gravitation constant	$6.67\times 10^{-11}~N-m^2/kg^2$	Earth's radius	$6.37\times 10^6~m$
Coulomb constant $(k_e)$	$8.99 \times 10^9 \frac{N-m^2}{C^2}$	Electron mass	$9.11\times 10^{-31}~kg$
Elementary charge $(e)$	$1.60\times 10^{-19}~C$	Proton/Neutron mass	$1.67\times 10^{-27}~kg$
Permittivity constant $(\epsilon_0)$	$8.85 \times 10^{-12} \frac{kg^2}{N-m^2}$	1.0  eV	$1.6\times 10^{-19}~J$
1 MeV	$10^6 \ eV$	atomic mass unit $(u)$	$1.66 \times 10^{-27} \ kg$
Planck's constant $(h)$	$6.626 \times 10^{-34} J - s$	Permittivity constant $(\epsilon_0)$	$8.85 \times 10^{-12} \frac{kg^2}{N-m^2}$
Planck's constant 2 ( $\hbar = h/2\pi$ )	$1.0546 \times 10^{-34} J - s$	R	8.31J/K - mole

## Physics 132-4 Equation Sheet, Final

$$\vec{F} = m\vec{a} = \frac{d\vec{p}}{dt}$$
  $a_c = \frac{v^2}{r}$   $KE = \frac{1}{2}mv^2$   $ME_0 = ME_1 = KE_1 + PE_1$   $\vec{p} = m\vec{v}$   $\vec{p}_0 = \vec{p}_1$ 

$$x = \frac{a}{2}t^2 + v_0t + x_0 \quad v = at + v_0 \qquad Q = C\Delta T = cm\Delta T = nC_v\Delta T \quad Q_{f,v} = mL_{f,v}$$

$$\Delta E_{int} = Q + W \quad W = \int \vec{F} \cdot d\vec{s} \to P\Delta V \quad \langle \vec{F} \rangle = \frac{\Delta \vec{p}}{\Delta t} \quad P = \frac{|\vec{F}|}{A} \quad PV = Nk_B T = nRT$$

$$\vec{I} = \int \vec{F} dt = \langle \vec{F} \rangle \Delta t = \Delta \vec{p} \quad \langle KE \rangle = \langle E_{kin} \rangle = \frac{1}{2} m \overline{v^2} \quad \langle E_{kin} \rangle = \frac{3}{2} k_B T \quad E_{int} = N \langle E_{kin} \rangle = \frac{3}{2} N k_B T$$

 $v_{rms} = \sqrt{\langle v^2 \rangle}$   $C_V = \frac{f}{2} N_A k_B$   $E_f = \frac{k_B T}{2}$   $E_{int} = \frac{f}{2} N k_B T$   $f \equiv$  number of degrees of freedom

$$E_{atom} = (n_x + n_y + n_z + \frac{3}{2})\epsilon_i \quad E = \sum_{i=1}^{3N} n_i\epsilon_i = q\epsilon_i \quad \Omega(N,q) = \frac{(q+3N-1)!}{q!(3N-1)!}$$
$$S = k_B \ln \Omega \quad \frac{1}{T} = \frac{dS}{dE} \quad q = \frac{E}{\hbar\omega_0} \quad C = \frac{1}{n}\frac{dE}{dT} \quad E = 3Nk_BT$$
$$\vec{F}_C = k_e \frac{q_1q_2}{r^2}\hat{r} \quad \vec{E} \equiv \frac{\vec{F}}{q_0} \quad \vec{E} = k_e \sum_i \frac{q_i}{r_i^2}\hat{r}_i \quad \vec{E} = k_e \int \frac{dq}{r^2}\hat{r}$$

$$\vec{E}_{plate} = 2\pi k_e \sigma \hat{k} \quad \vec{E}_{ring} = k_e \frac{qz}{(z^2 + R^2)^{3/2}} \hat{k} \quad \Delta V \equiv \frac{\Delta PE}{q_0} = -\int_A^B \vec{E} \cdot d\vec{s} \quad V = k_e \frac{q}{r} \quad PE = qV$$

$$V = k_e \sum_n \frac{q_n}{r_n} \quad V = k_e \int \frac{dq}{r} \quad V = Ed \quad E_x = -\frac{\partial V}{\partial x} \quad E_y = -\frac{\partial V}{\partial y} \quad E_z = -\frac{\partial V}{\partial z}$$

$$I \equiv \frac{dQ}{dt} \quad V = IR \quad P = IV \quad R_{equiv} = \sum R_i \quad \frac{1}{R_{equiv}} = \sum \frac{1}{R_i}$$

The algebraic sum of the potential changes across all the elements of a closed loop is zero.

The sum of the currents entering a junction is equal to the sum of the currents leaving the junction.

$$I = nev_d A \quad v_d = \frac{eE\tau}{m_e} \quad \tau = \frac{l_{avg}}{v_{avg}} \quad l_{avg} = \frac{1}{\sqrt[3]{n}} \quad \langle E \rangle = \frac{1}{2}mv_{rms}^2 = \frac{3}{2}k_B T$$

$$\vec{F}_B = q\vec{v} \times \vec{B} \quad |\vec{F}_B| = |qvB\sin\theta| \quad \vec{F}_c = -m\frac{v^2}{r}\hat{r} \qquad \frac{dN}{dt} = -\lambda t \quad N = N_0 e^{-\lambda t} \quad t_{1/2} = \frac{\ln 2}{\lambda}$$
$$y = A\cos(kx - \omega t) \quad k\lambda = 2\pi = \omega T \quad \frac{\lambda}{T} = c \quad f = \frac{1}{T}$$

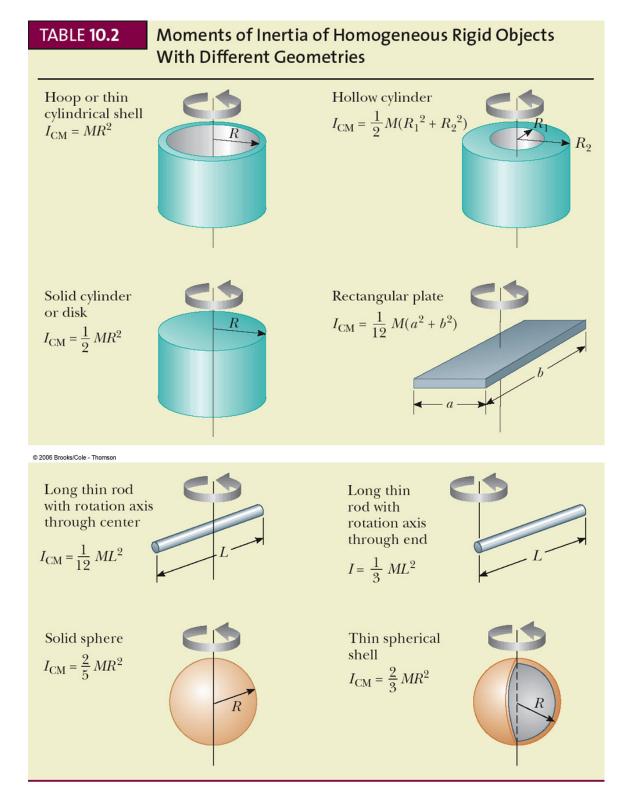
 $E = E_m \sin(kx - \omega t) \quad B = B_m \sin(kx - \omega t) \quad \vec{S} = \frac{1}{\mu_0} \vec{E} \times \vec{B} \quad |\vec{S}| = I = \frac{E^2}{\mu_0 c} \quad \frac{E_m}{B_m} = c$ 

$$I = 4I_0 \cos^2\left(\frac{\pi d}{\lambda}\sin\theta\right) \quad I = I_m \left[\frac{\sin\left(\frac{\pi a}{\lambda}\sin\theta\right)}{\frac{\pi a}{\lambda}\sin\theta}\right]^2 \quad I = I_m \cos^2\left(\frac{\pi d}{\lambda}\sin\theta\right) \left[\frac{\sin\left(\frac{\pi a}{\lambda}\sin\theta\right)}{\frac{\pi a}{\lambda}\sin\theta}\right]^2$$
$$\delta = d\sin\theta = m\lambda \quad \delta = a\sin\theta = m\lambda \quad \phi = k\delta \quad \sin\theta_R = \frac{\lambda}{a}$$

$$\begin{split} L &= I\omega = mv_{l}r \quad I \sum m_{i}r_{i}^{2} \quad I = I_{cm} + mR^{2} \quad L_{0} = L_{1} \quad E = \frac{1}{2}mv_{r}^{2} + \frac{L^{2}}{mr^{2}} - k_{e}\frac{e^{2}}{r} \\ &= \frac{1}{\lambda} = R_{H}\left(\frac{1}{n_{f}^{2}} - \frac{1}{n_{i}^{2}}\right) \quad E_{n} = -\frac{13.6 \ eV}{n^{2}} \quad E = hf = h\frac{c}{\lambda} \\ &- \frac{h^{2}}{2m}\left(\frac{d^{2}}{dr^{2}}\right)\Psi(r) + \frac{L^{2}}{2mr^{2}}\Psi(r) + V\Psi(r) = E\Psi(r) \\ &= \frac{df(x)}{dx} = \lim_{\Delta x \to 0} \frac{f(x + \Delta x) - f(x)}{\Delta x} \quad \frac{dx^{n}}{dx} = nx^{n-1} \quad \frac{de^{x}}{dx} = e^{x} \quad \frac{df(u)}{dx} = \frac{df}{du}\frac{du}{dx} \\ &= \frac{d}{dx}f(x) \cdot g(x) = f\frac{dg}{dx} + g\frac{df}{dx} \quad \frac{d\ln x}{dx} = \frac{1}{x} \quad \frac{d}{dx}\cos ax = -a\sin ax \quad \frac{d}{dx}\sin ax = a\cos ax \\ &\int_{a}^{b} f(x)dx = \lim_{\Delta x \to 0} \sum_{n=1}^{N} f(x)\Delta x \quad \int x^{n}dx = \frac{x^{n+1}}{n+1} \quad \int \sin^{2}(ax)dx = \frac{x}{2} - \frac{\sin(ax)}{4a} \\ &\int e^{x}dx = e^{x} \quad \int \frac{1}{x}dx = \ln x \int \frac{1}{\sqrt{x^{2} + a^{2}}}dx = \ln \left[x + \sqrt{x^{2} + a^{2}}\right] \quad \int \frac{x}{\sqrt{x^{2} + a^{2}}}dx = \sqrt{x^{2} + a^{2}} \\ &\int \frac{x^{2}}{\sqrt{x^{2} + a^{2}}}dx = \frac{1}{2}x\sqrt{x^{2} + a^{2}} - \frac{1}{2}a^{2}\ln \left[x + \sqrt{x^{2} + a^{2}}\right] \quad \int \frac{x^{3}}{\sqrt{x^{2} + a^{2}}}dx = \frac{1}{3}(-2a^{2} + x^{2})\sqrt{x^{2} + a^{2}} \end{split}$$

$$\langle x \rangle = \frac{1}{N} \sum_{i} x_{i} \quad \sigma = \sqrt{\frac{\sum_{i} (x_{i} - \langle x \rangle)^{2}}{N - 1}} \quad N = \frac{b - a}{\Delta x} \quad A = 4\pi r^{2} \quad V = Ah \quad V = \frac{4}{3}\pi r^{3}$$

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hydrogen 1	-		1	0777.	-	e	184	ē	10	10	515).		1818) (1	•••	6.53		66 V	² He
1.0079 lithium	beryllium	1										Ĩ	boron	carbon	nitrogen	oxygen	fluorine	4.0026 neon
3	4												5	6	7	8	9	10
Li	Be												B	C	N	0	F	Ne
6.941	9.0122												10.811	12.011	14.007	15.999	18.998	20.180
sodium 11	magnesium 12												aluminium 13	silicon 14	phosphorus 15	sulfur 16	chlorine 17	argon 18
	100000													2.2		2.5		
Na	Mg												AI	Si	Ρ	S	CI	Ar
22.990 potassium	24.305 calcium		a second base	titanium	vanadium	a hana an isana		lana		nickel		-	26.982 gallium	28.086	30.974	32.065 selenium	35.453 bromine	39.948
19	20		scandium 21	22	23	chromium 24	manganese 25	iron 26	cobalt 27	28	copper 29	zinc 30	31	germanium 32	arsenic 33	34	35	krypton 36
K	Ca		Co	100	1	0	D. //	<b>1</b>	0	N 1 5	0		0	0		0	D	17
	Ja		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
39.098	40.078		44.956	47.867	50.942	51.996	54.938	55.845	58.933	58.693	63.546	65.39	69.723	72.61	74.922	78.96	79.904	83.80
39.098 rubidium	40.078 strontium		44.956 yttrium	47.867 zirconium	50.942 niobium	51.996 molybdenum	54.938 technetium	55.845 ruthenium	58.933 rhodium	58.693 palladium	silver	65.39 cadmium	69.723 indium	72.61 tin	74.922 antimony	78.96 tellurium	79.904 iodine	83.80 xenon
39.098 rubidium <b>37</b>	40.078 strontium <b>38</b>		44.956 yttrium <b>39</b>	47.867 zirconium <b>40</b>	50.942 niobium <b>41</b>	51.996 molybdenum <b>42</b>	54.938 technetium <b>43</b>	55.845 ruthenium <b>44</b>	58.933 rhodium <b>45</b>	58.693 palladium <b>46</b>	silver 47	65.39 cadmium <b>48</b>	69.723 indium <b>49</b>	72.61 tin <b>50</b>	74.922 antimony <b>51</b>	78.96 tellurium <b>52</b>	79.904	83.80 xenon <b>54</b>
39.098 rubidium 37 <b>Rb</b>	40.078 strontium 38 Sr		44.956 yttrium <b>39</b> <b>Y</b>	47.867 zirconium 40 <b>Zr</b>	50.942 niobium 41 <b>Nb</b>	51.996 molybdenum 42 MO	54.938 technetium 43 <b>TC</b>	<sup>55.845</sup> ruthenium 44 <b>Ru</b>	58.933 rhodium 45 <b>Rh</b>	58.693 palladium 46 Pd	47 Ag	65.39 cadmium 48 <b>Cd</b>	69.723 indium 49 In	72.61 tin 50 <b>Sn</b>	74.922 antimony 51 <b>Sb</b>	78.96 tellurium 52 <b>Te</b>	79.904 iodine 53	83.80 xenon 54 Xe
39.098 rubidium 37 <b>Rb</b> 85.468	40.078 strontium 38 Sr 87.62		44.956 yttrium 39 Y 88.906	47.867 zirconium 40 <b>Zr</b> 91.224	50.942 niobium 41 Nb 92.906	51.996 molybdenum 42 Mo 95.94	54.938 technetium 43 <b>TC</b> [98]	55.845 ruthenium 44 Ru 101.07	58.933 rhodium 45 Rh 102.91	58.693 palladium 46 Pd 106.42	47 47 Ag 107.87	65.39 cadmium 48 <b>Cd</b> 112.41	69.723 indium 49 In 114.82	72.61 tin 50 <b>Sn</b> 118.71	74.922 antimony 51 Sb 121.76	78.96 tellurium 52 Te 127.60	79.904 iodine 53 126.90	83.80 xenon 54 Xe 131.29
39.098 rubidium 37 <b>Rb</b>	40.078 strontium 38 Sr	57-70	44.956 yttrium <b>39</b> <b>Y</b>	47.867 zirconium 40 <b>Zr</b>	50.942 niobium 41 <b>Nb</b>	51.996 molybdenum 42 MO	54.938 technetium 43 <b>TC</b>	<sup>55.845</sup> ruthenium 44 <b>Ru</b>	58.933 rhodium 45 <b>Rh</b>	58.693 palladium 46 Pd	47 Ag	65.39 cadmium 48 <b>Cd</b>	69.723 indium 49 In	72.61 tin 50 <b>Sn</b>	74.922 antimony 51 <b>Sb</b>	78.96 tellurium 52 <b>Te</b>	79.904 iodine 53	83.80 xenon 54 Xe
39.098 rubidium 37 Rb 85.468 caesium	40.078 strontium 38 Sr 87.62 barium	57-70 ★	44.956 yttrium <b>39</b> <b>Y</b> 88.906 lutetium	47.867 zirconium 40 Zr 91.224 hafnium	50.942 niobium 41 Nb 92.906 tantalum	51.996 molybdenum 42 Mo 95.94 tungsten	54.938 technetium 43 TC [98] rhenium	55.845 ruthenium 44 Ru 101.07 osmium	58.933 rhodium 45 <b>Rh</b> 102.91 iridium	58.693 palladium 46 Pd 106.42 platinum	47 47 Ag 107.87 gold	65.39 cadmium 48 Cd 112.41 mercury 80	69.723 indium 49 In 114.82 thallium	72.61 tin 50 Sn 118.71 lead	74.922 antimony 51 <b>Sb</b> 121.76 bismuth	78.96 tellurium 52 Te 127.60 polonium	79.904 iodine 53 126.90 astatine	83.80 xenon 54 Xe 131.29 radon
39.098 rubidium 37 <b>Rb</b> 85.468 caesium 55 <b>CS</b> 132.91	40.078 strontium 38 Sr 87.62 barium 56 Ba 137.33		44.956 yttrium 39 Y 88.906 lutetium 71 Lu 174.97	47.867 zirconium 40 <b>Zr</b> 91.224 hafnium 72 Hff 178.49	50.942 niobium 41 Nb 92.906 tantalum 73 Ta 180.95	51.996 molybdenum 42 Mo 95.94 tungsten 74 W 183.84	54.938 technetium 43 <b>TC</b> [98] menium 75 <b>Re</b> 186.21	55.845 ruthenium 44 Ruu 101.07 osmium 76 OS 190.23	58.933 rhodium 45 <b>Rh</b> 102.91 iridium 77 <b>Ir</b> 192.22	58.693 palladium 46 Pd 106.42 platinum 78 Pt 195.08	silver 47 Ag 107.87 gold 79 Au 196.97	65.39 cadmium 48 Cd 112.41 mercury 80 Hg 200.59	69.723 indium 49 In 114.82 thallium	72.61 tin 50 Sn 118.71 lead 82 Pb 207.2	74.922 antimony 51 <b>Sb</b> 121.76 bismuth 83	78.96 teilurium 52 Te 127.60 polonium 84	79.904 lodine 53 126.90 astatine 85	83.80 xenon 54 Xe 131.29 radon 86
39.098 rubidium 37 <b>Rb</b> 85.468 caesium 55 <b>Cs</b>	40.078 strontium 38 <b>Sr</b> 87.62 barium 56 <b>Ba</b>		44,956 yttrium 39 Y 88,906 Iutetium 71 Lu	47.867 zirconium 40 <b>Zr</b> 91.224 hafnium 72 Hf	50.942 niobium 41 Nb 92.906 tantalum 73 Ta	51.996 molybdenum 42 Mo 95.94 tungsten 74 W	54,938 technetium 43 <b>TC</b> [98] thenium 75 <b>Re</b>	55.845 ruthenium 44 Ru 101.07 osmium 76 OS	58.933 rhodium 45 <b>Rh</b> 102.91 iridium 77 <b>Ir</b>	58.693 palladium 46 Pd 106.42 platinum 78 Pt	silver 47 Ag 107.87 gold 79 Au	65.39 cadmium 48 Cd 112.41 mercury 80 Hg	69,723 indium 49 <u>In</u> 114.82 thallium 81 <b>TI</b>	72.61 tin 50 Sn 118.71 lead 82 Pb	74.922 antimony 51 Sb 121.76 bismuth 83 Bi	78.96 tellurium 52 Te 127.60 polonium 84 PO	79.904 iodine 53 126.90 astatine 85 At	83.80 xenon 54 Xe 131.29 radon 86 Rn
39.098 rubidium 37 <b>Rb</b> 85.468 caesium 55 <b>CS</b> 132.91 francium 87	40.078 strontium 38 Sr 87.62 barium 56 Baa 137.33 radium 88	* 89-102	44.956 yttrium 39 Y 88.906 lutetium 71 Luu 174.97 lawrencium 103	47.867 zirconium 40 <b>Zr</b> 91.224 hafnium 72 <b>Hff</b> 178.49 rutherfordium 104	50.942 niobium 41 Nb 92.906 tantalum 73 Ta 180.95 dubnium 105	51.996 molybdenum 42 Mo 95.94 tungsten 74 W 183.84 seaborglum 106	54.938 technetium 43 TC 198 98 thenium 75 Re 186.21 bohrium 107	55.845 ruthenium 44 Ruu 101.07 osmium 76 OS 190.23 hassium 108	58.933 rhodium 45 Rh 102.91 iridium 77 Ir 192.22 meltnerium 109	58.693 palladium 46 Pd 106.42 platinum 78 Pt 195.08 ununnilium 110	silver 47 Ag 107.87 gold 79 Au 196.97 unununium 111	65.39 cadmium 48 Cd 112.41 mercury 80 Hg 200.59 ununbium 112	69,723 indium 49 <u>In</u> 114.82 thallium 81 <b>TI</b>	72.61 tin 50 Sn 118.71 lead 82 Pb 207.2 ununquadium 114	74.922 antimony 51 Sb 121.76 bismuth 83 Bi	78.96 tellurium 52 Te 127.60 polonium 84 PO	79.904 iodine 53 126.90 astatine 85 At	83.80 xenon 54 Xe 131.29 radon 86 Rn
39.098 rubidium 37 <b>Rb</b> 85.468 caesium 55 <b>CS</b> 132.91 francium	40.078 strontium 38 Sr 87.62 barium 56 Ba 137.33 radium	*	44.956 yttrium 39 Y 88.906 lutetium 71 Luu 174.97 lawrencium	47.867 zirconium 40 Zr 91.224 hafnium 72 Hff 178.49 rutherfordium	50.942 niobium 41 <b>Nbb</b> 92.906 tantalum 73 <b>Ta</b> 180.95 dubnlum	51.996 molybdenum 42 Mo 95.94 tungsten 74 W 183.84 seaborglum	54.938 technetium 43 TC 198 98 thenium 75 <b>Re</b> 186.21 bohrium	55.845 ruthenium 44 Ruu 101.07 osmium 76 OS 190.23 hassium	58,933 rhodium 45 Rh 102,91 iridium 77 Ir 192,22 meitnerium	58.693 palladium 46 Pd 106.42 platinum 78 Pt 195.08 ununnilium 110	silver 47 Ag 107.87 gold 79 Au 196.97 unununum	65.39 cadmium 48 Cd 112.41 mercury 80 Hg 200.59 ununbium 112	69,723 indium 49 <u>In</u> 114.82 thallium 81 <b>TI</b>	72.61 tin 50 Sn 118.71 lead 82 Pb 207.2 ununquadium	74.922 antimony 51 Sb 121.76 bismuth 83 Bi	78.96 tellurium 52 Te 127.60 polonium 84 PO	79.904 iodine 53 126.90 astatine 85 At	83.80 xenon 54 Xe 131.29 radon 86 Rn

*Lanthanide series	lanthanum 57	cerium 58	praseodymium 59	neodymium 60	promethium 61	samarium 62	europium 63	gadolinium 64	terbium 65	dysprosium 66	holmium 67	erbium 68	thulium 69	ytterbium 70
Lanthaniue series	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dv	Ho	Er	Tm	Yb
	138.91	140.12	140.91	144.24	[145]	150.36	151.96	157.25	158.93	162.50	164.93	167.26	168.93	173.04
	actinium	thorium	protactinium	uranium	neptunium	plutonium	americium	curium	berkelium	californium	einsteinium	fermium	mendelevium	nobelium
* * Actinide series	89	90	91	92	93	94	95	96	97	98	99	100	101	102
,	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
	[227]	232.04	231.04	238.03	[237]	[244]	[243]	[247]	[247]	[251]	[252]	[257]	[258]	[259]