What is the Energy of an Electron?

The Coulomb force binds an electron and a proton into a hydrogen atom with a force that is mathematically identical to the gravitational force that binds the planets in our Solar System, the Moon to the Earth, *etc.* What is the energy E_e of an electron?



• The Organizing Principle.

$$ME_{0} = ME_{1}$$

$$KE_{0} + PE_{0} = ME_{1} + PE_{1}$$

$$\frac{1}{2}mv_{0}^{2} + PE_{0} = \frac{1}{2}mv_{1}^{2} + PE_{1}$$

• The Forces

$$\vec{F}_{grav} = -rac{Gm_1m_2}{r_{12}^2}\hat{r}_{12}$$
 $\vec{F}_{coul} = rac{k_eq_1q_2}{r_{12}^2}\hat{r}_{12}$

The simulation is here.

Atomic Spectroscopy - 1

Light from a hydrogen spectrum tube is incident on a diffraction grating in a spectrometer. A narrow, red line appears at $\theta_1 = 20.50^{\circ}$. The grating has a line density of 13,400 lines/inch. What is the wavelength of the light? What is the energy of the photons?



The Diffraction Grating













An electron beam strikes a gas of hydrogen atoms.

- What is the minimum speed the electrons must have to cause the emission of $\lambda = 656 \text{ nm}$ light from the 3 \rightarrow 2 transition of hydrogen?
- What is the electric potential difference the electrons must fall through to be accelerated to this speed?

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The Kinetic Energy in Polar Coordinates - 1 14



The Kinetic Energy in Polar Coordinates - 1 15



Orbits

A Russian Artica satellite that monitors polar weather follows an elliptical orbit around the Earth at an altitude of $h = 300 \ km$ above the surface (radius $r_s = 6.67 \times 10^6 \ m$). At one point in its orbit its velocity is measured to be

$$\vec{v} = 4.1 \times 10^3 \ m/s \ \hat{r} + 7.5 \times 10^3 \ m/s \ \hat{ heta}$$

What is the angular momentum? What is the total energy? What is the distance of closest approach to the Earth? The satellite mass is $m_s = 600 \ kg$.



$$\begin{array}{l} R_{earth} = 6.37 \times 10^6 \ m \\ m_{earth} = 5.97 \times 10^{24} \ kg \\ G = 6.673 \times 10^{-11} \ Nm^2/kg^2 \end{array}$$



