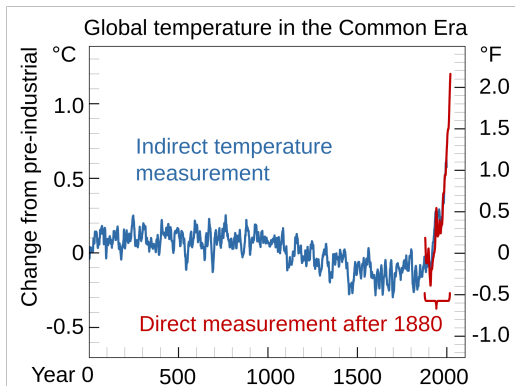


Understanding climate change is driven, in part, by our understanding of temperatures from deep time (*i.e.*, long ago). Below is a global surface temperature reconstruction over the last 2000 years using proxy data from tree rings, corals, and ice cores in blue [1]. Directly observed data is in red [2].

1. R. Neukom *et al.*, *Nature Geoscience*. 12 (8): 643–649.
2. V. Masson-Delmotte *et al.*, *IPCC AR5 WG1 2013*. pp. 383–464.

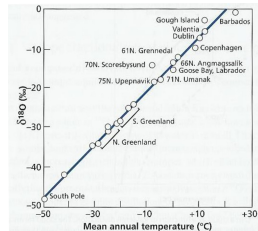
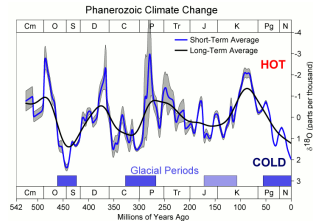
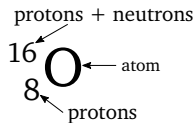


How do we measure temperatures in deep time? 2

- 1 Nuclear isotopes can be used as thermometers.
- 2 Vapor pressure of H_2^{16}O ('normal' isotope of oxygen) is 1% higher than vapor pressure of H_2^{18}O (replace ^{16}O with ^{18}O) which has two more neutrons.
- 3 Evaporation creates vapor poorer in the heavier heavier isotope ^{18}O leaving the remaining water enriched in ^{18}O .
- 4 Water vapor is carried over polar regions where the H_2^{18}O molecules condense more readily than H_2^{16}O so rain is enriched in the heavy isotope and vapor is depleted even more. The temperature proxy is

$$\delta^{18}\text{O} = \left(\frac{^{18}\text{O}/^{16}\text{O}_{\text{measured}}}{^{18}\text{O}/^{16}\text{O}_{\text{standard}}} - 1 \right) \times 1000$$

- 5 The plot shows the $\delta^{18}\text{O}$ - T connection.



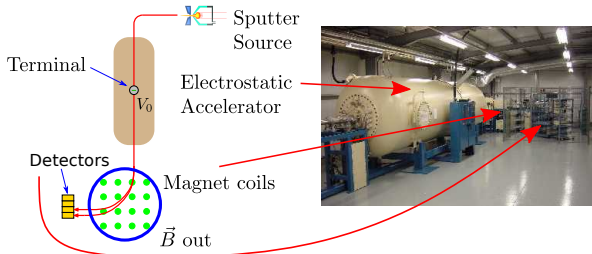
- 1 The temperature proxy is

$$\delta^{18}\text{O} = \left(\frac{{}^{18}\text{O}/{}^{16}\text{O}_{\text{measured}}}{{}^{18}\text{O}/{}^{16}\text{O}_{\text{standard}}} - 1 \right) \times 1000$$

where ${}^{18}\text{O}$ and ${}^{16}\text{O}$ are the amounts of different oxygen isotopes extracted from ice or ocean floor drilling cores.

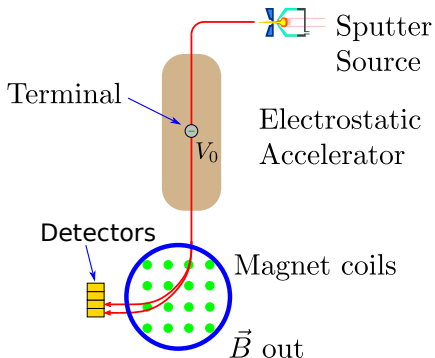
- 2 The chemistry of the two oxygen isotopes is very similar making it difficult to precisely separate them.
- 3 The masses of the two isotopes are significantly different so use electric and magnetic forces to do the separation.

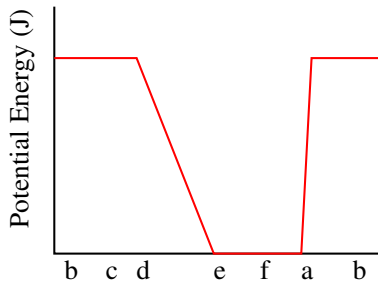
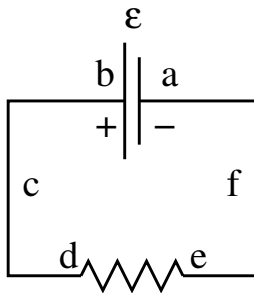
A Mass Spectrometer

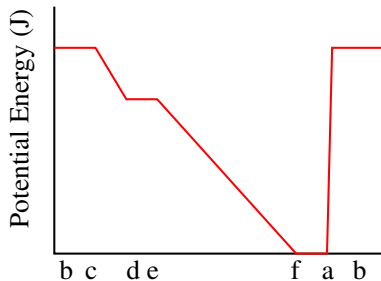
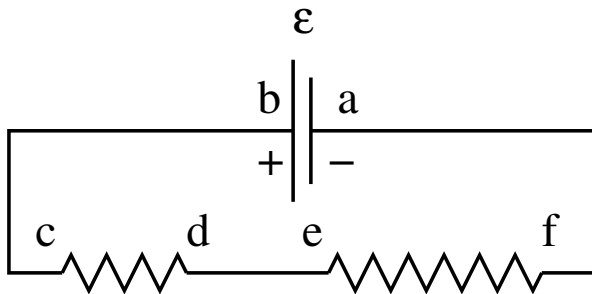


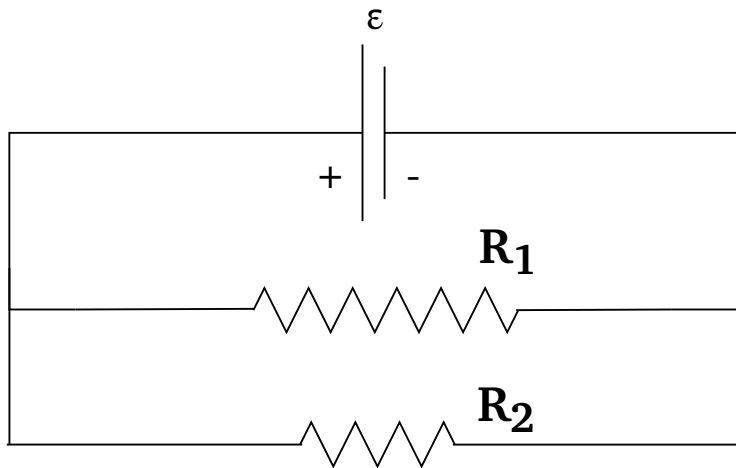
Measuring Temperature a Million Years Ago 4

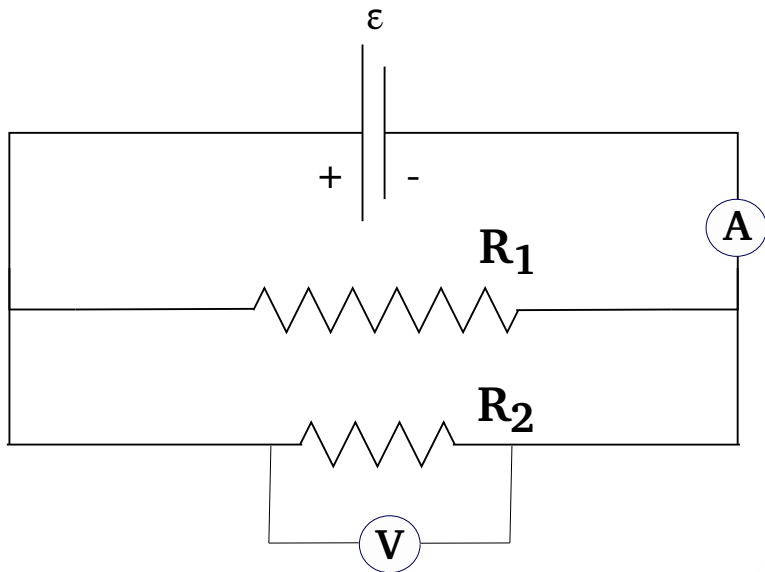
Isotopes of oxygen, ^{16}O and ^{18}O , are accelerated twice across a potential difference $V_0 = 5.0 \times 10^6 \text{ V}$. Each atom carries an extra electron initially, but that electron and three others are stripped off the oxygen in the accelerator by a thin foil at the terminal. The oxygen has a charge $+3e$ and is moving horizontally when it enters a uniform magnetic field pointing straight up with $B = 3.0 \text{ T}$. The positively-charged atoms follow a circular path before striking detectors that measure their position. The variation in the trajectory of each atom leads to an uncertainty in their final position of about $\pm 0.007 \text{ m}$. Will the spectrometer be able to separate the isotopes?





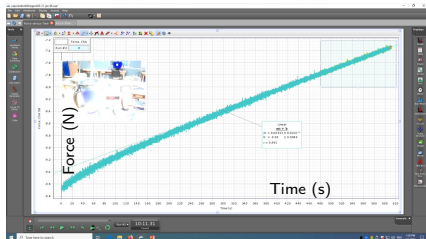




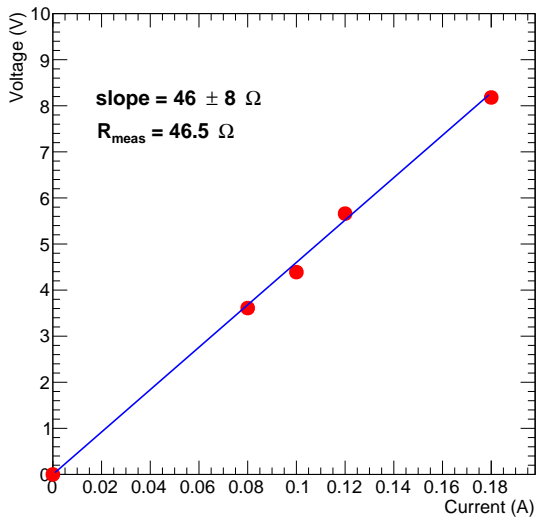


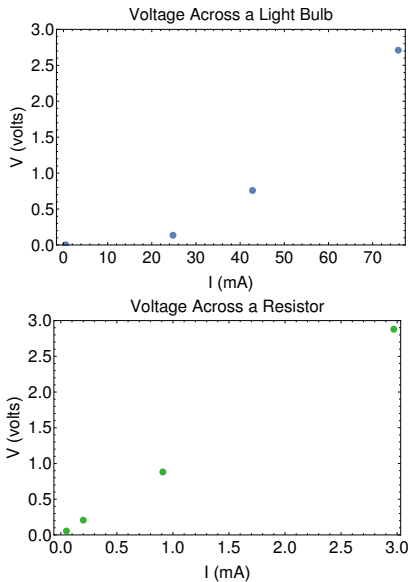
In our study of thermodynamics you measured the latent heat of vaporization L_V of liquid nitrogen by heating an open container of liquid nitrogen with a resistor with current \mathcal{I} and voltage \mathcal{V} . See the figure below. The electrical power generated in the resistor by the current heated the liquid nitrogen by a known amount. What is the relationship between power and the properties of the circuit \mathcal{I} and \mathcal{V} ? Using the values below compare the power dissipated in the circuit with the thermal energy needed to evaporate the liquid nitrogen. Do they agree?

slope (current on)	$0.0035 \pm 0.0002 \text{ N/s}$
slope (current off)	$0.0012 \pm 0.0003 \text{ N/s}$
L_V (accepted)	$1.99 \times 10^5 \text{ J/kg}$
current \mathcal{I}	2.01 A
voltage \mathcal{V}	23.23 V

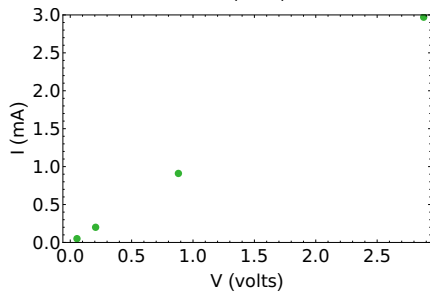
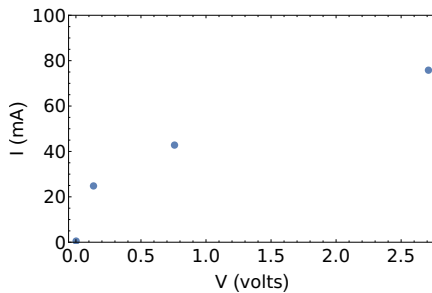


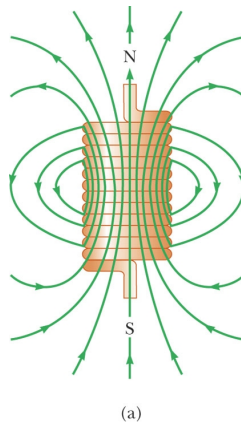
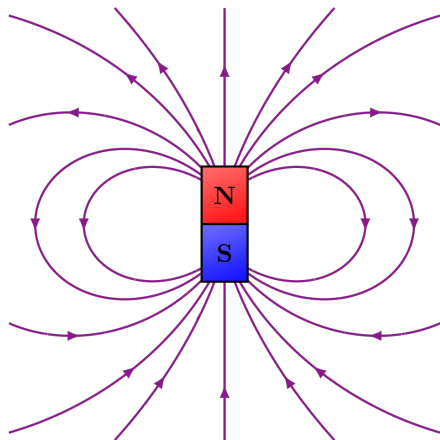
Ohm's Law





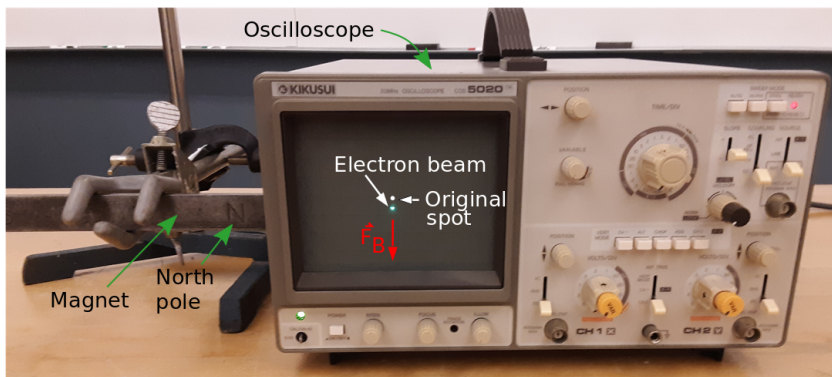
Note difference in horizontal scales.



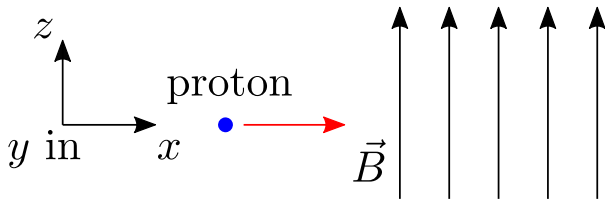


© 2006 Brooks/Cole - Thomson

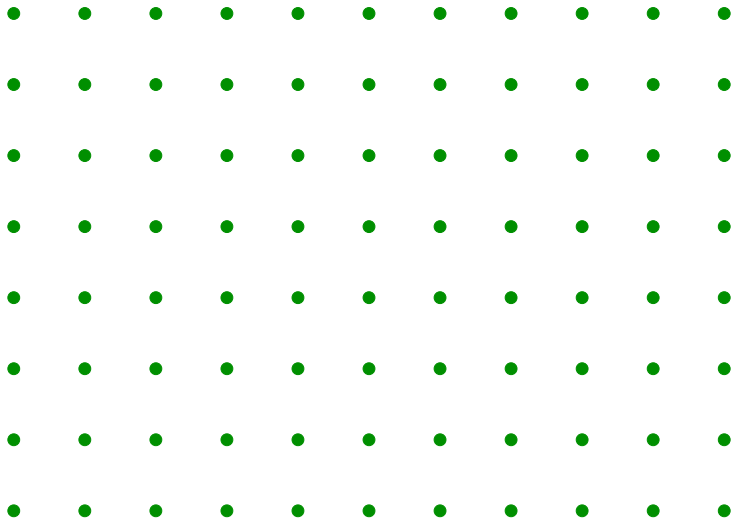
Magnetic Force on Moving Charged Particles 14



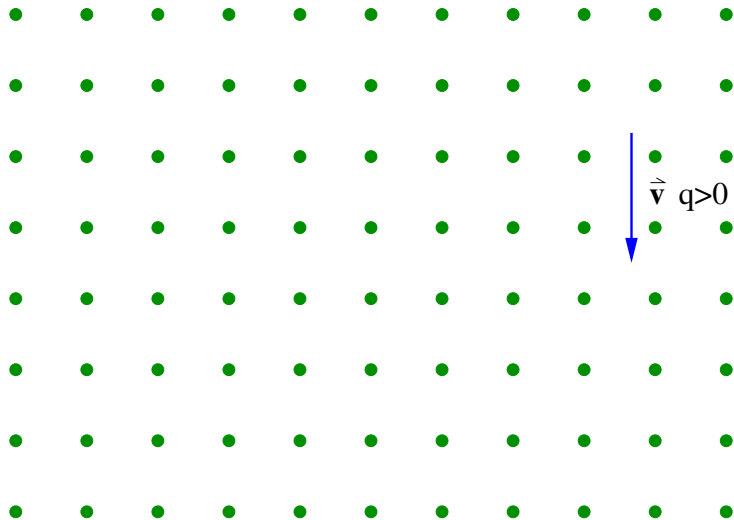
A uniform magnetic field has a magnitude $|\vec{B}| = 1.2 \text{ T}$ and points straight up. A proton with velocity $\vec{v} = 3.2 \times 10^7 \hat{i} \text{ m/s}$ enters the field moving horizontally. What is the magnitude and direction of the force on the proton? How would the force change for an electron moving with the same initial velocity? Describe the trajectory of the particle.



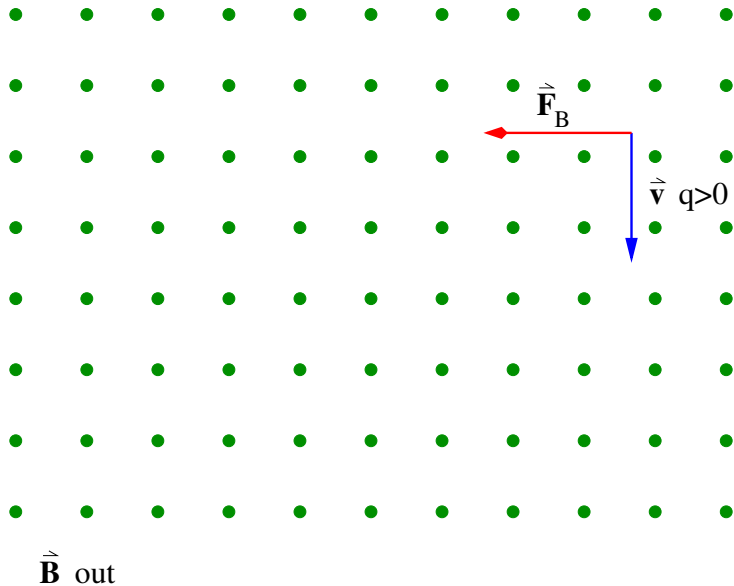
Demo is [here](#).

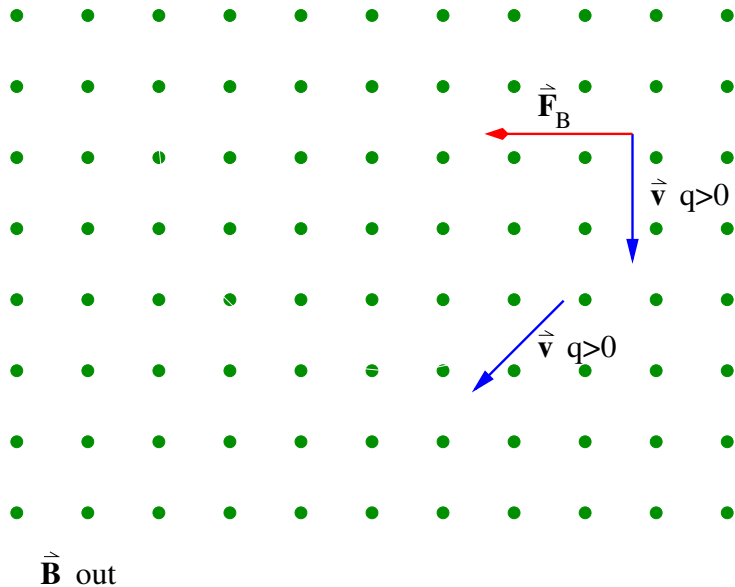


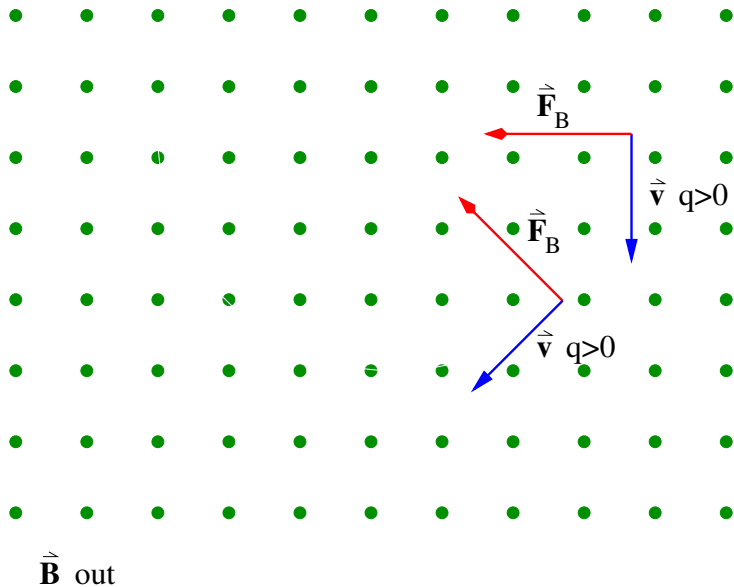
$\vec{\mathbf{B}}$ out

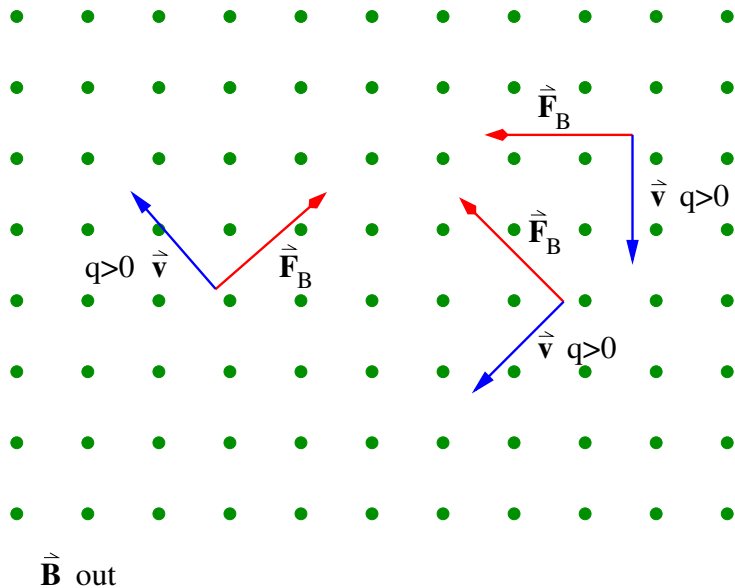


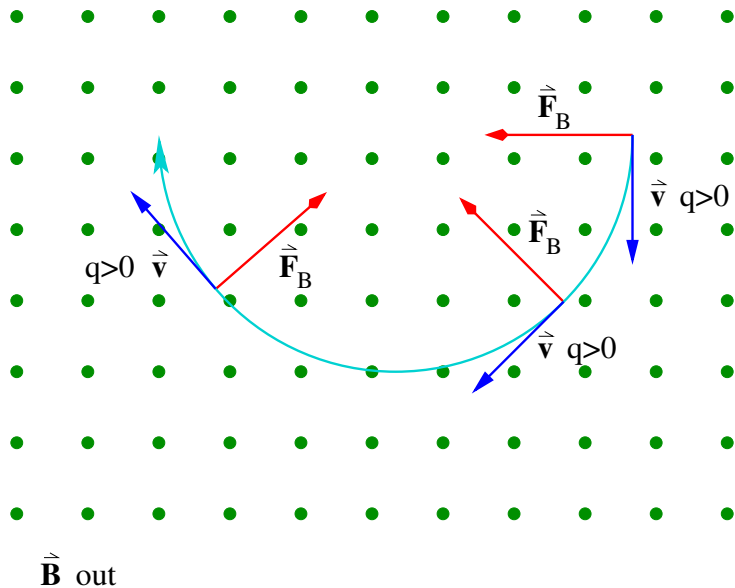
$\vec{\mathbf{B}}$ out



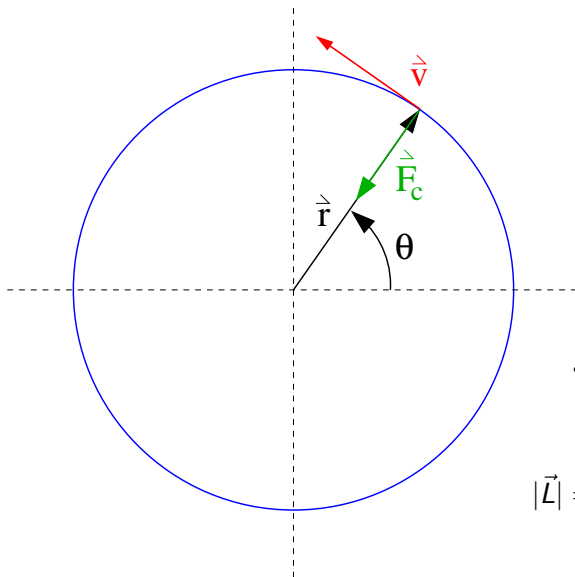








Circular Motion and Centripetal Force Summary 23



$$\vec{F}_c \perp \vec{v} \quad \vec{F}_c \parallel \vec{r}$$

$$\vec{a}_c \perp \vec{v} \quad \vec{a}_c \parallel \vec{r}$$

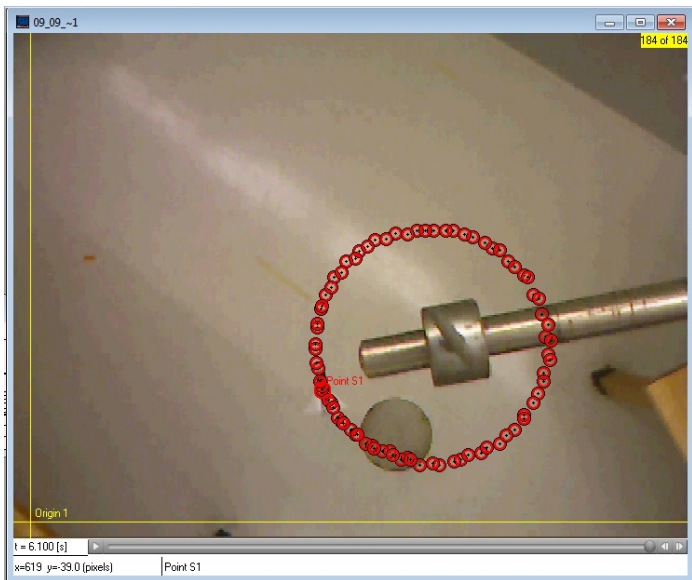
$$\vec{v} \perp \vec{r}$$

$$a_c = \frac{v^2}{r} \quad |\vec{F}_c| = \frac{mv^2}{r}$$

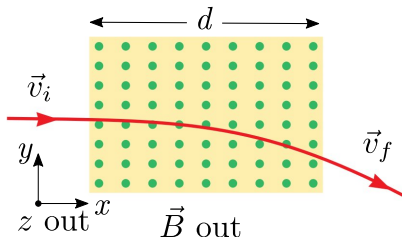
$$|\vec{L}| = mvr = I_L \omega \quad I_L = \sum mr_i^2$$

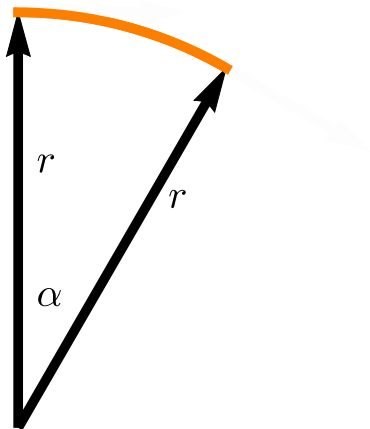
$$C = 2\pi r$$

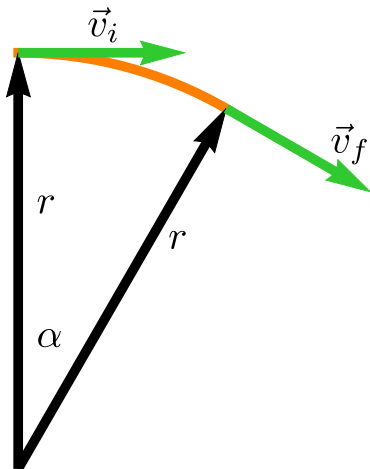


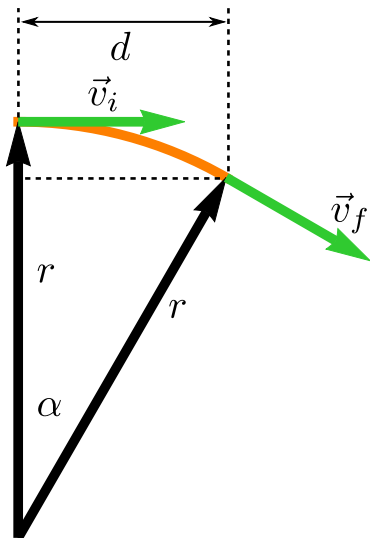


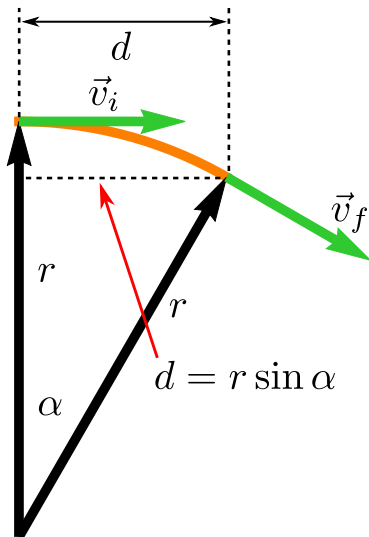
Protons having an initial speed of $v_i = 3.1 \times 10^7 \text{ m/s}$ are moving in the positive x direction and enter a magnetic field $\vec{B} = 0.05 \hat{k} \text{ T}$ directed out of the plane of the page and extending a horizontal distance $d = 1.0 \text{ m}$ as shown in the figure. (a) What is the radius of the arc the proton follows? (b) Find the angle α between the initial velocity vector \vec{v}_i of the proton beam and the final velocity vector \vec{v}_f after the beam emerges from the field. Ignore relativistic effects and note that $1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$.





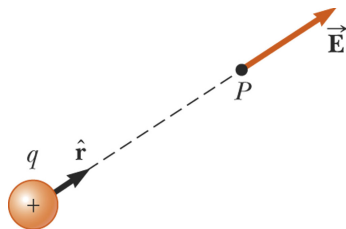






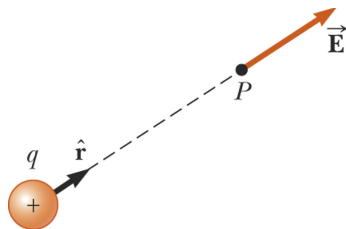
Coulomb's Law

$$d\vec{E} = k_e \frac{dq \hat{r}}{r^2} = \frac{1}{4\pi\epsilon_0} \frac{dq \hat{r}}{r^2}$$



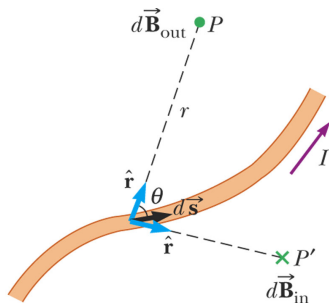
Coulomb's Law

$$d\vec{E} = k_e \frac{dq\hat{r}}{r^2} = \frac{1}{4\pi\epsilon_0} \frac{dq\hat{r}}{r^2}$$

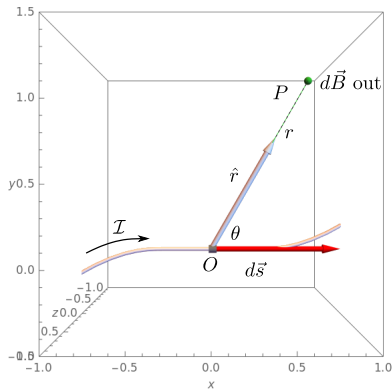
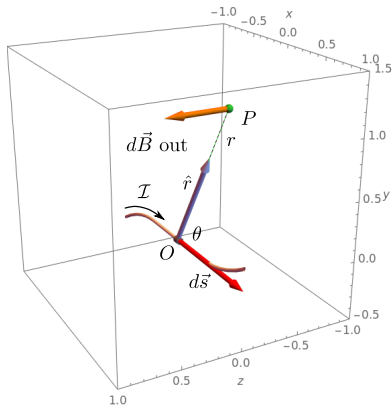


Biot-Savart Law

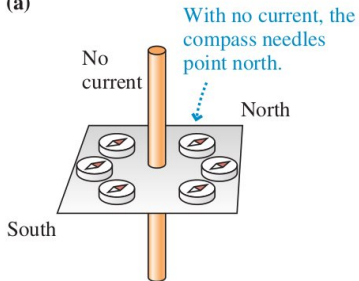
$$d\vec{B} = k_m \frac{I d\vec{s} \times \hat{r}}{r^2} = \frac{\mu_0}{4\pi} \frac{I d\vec{s} \times \hat{r}}{r^2}$$



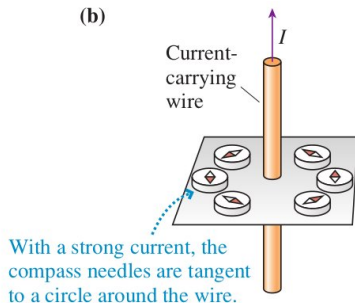
The figures show two views of a segment of a current. What is the magnetic field $d\vec{B}$ at the point P due to the infinitesimal chunk of current at O in terms of the current \mathcal{I} , $d\vec{s}$, r , and the angle θ ?



(a)



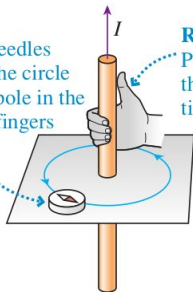
(b)



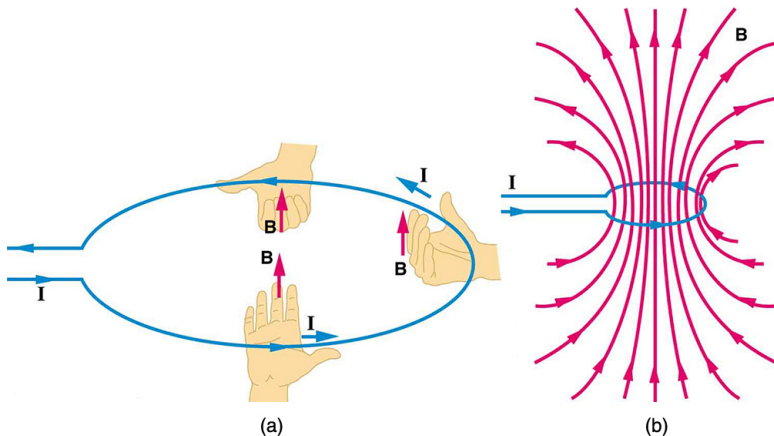
(c)

The compass needles are tangent to the circle with the north pole in the direction your fingers are pointing.

Right-hand rule:
Point your right thumb in the direction of the current.

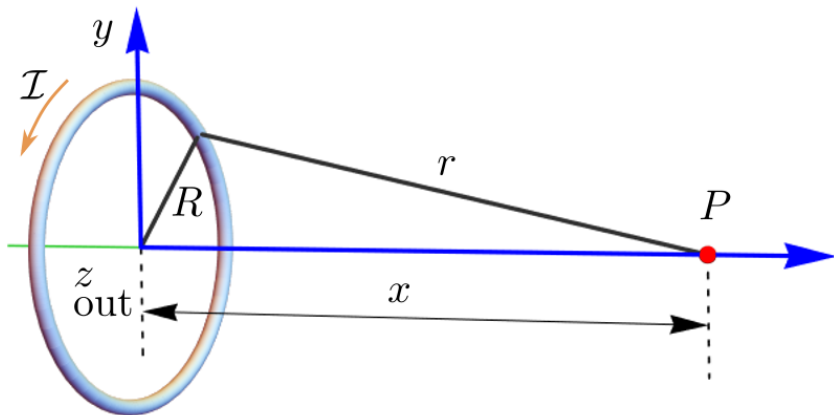


Demo is [here](#).

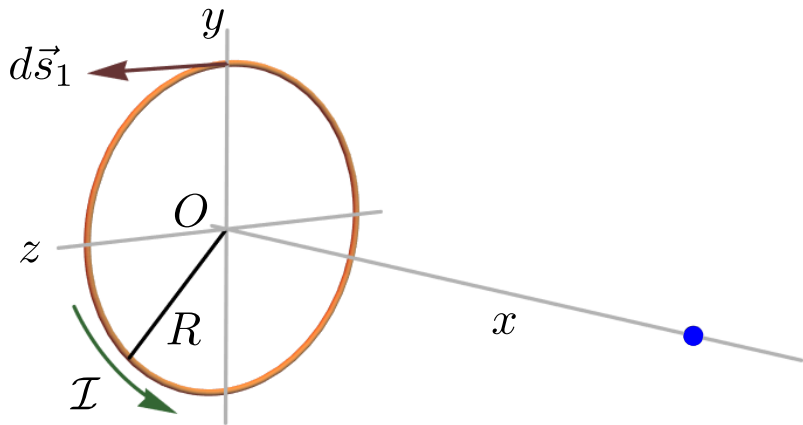


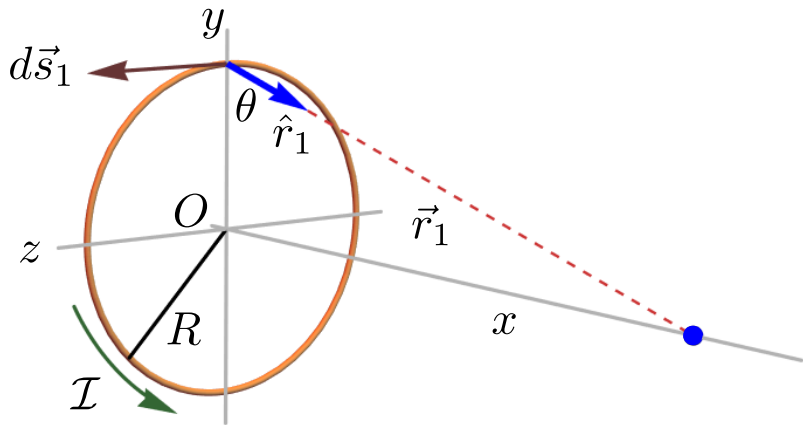
Demo is [here](#).

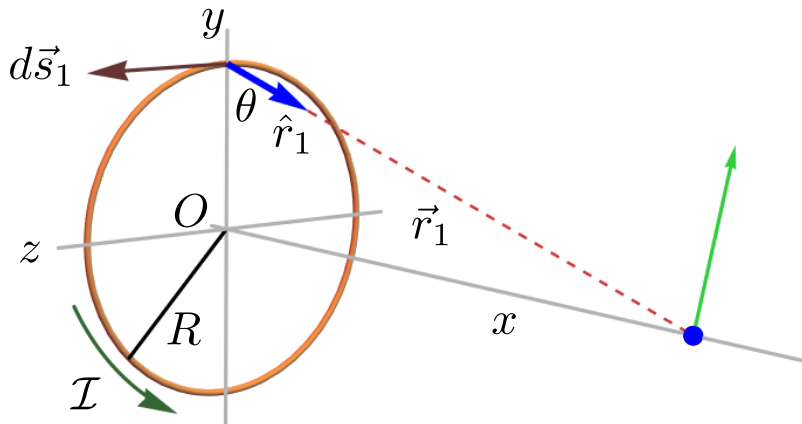
A ring of radius R as shown in the figure has a current \mathcal{I} . Calculate the magnetic field \vec{B} along the axis of the ring at a point lying a distance x from the center of the ring. Get your answer in terms of R , x , \mathcal{I} .

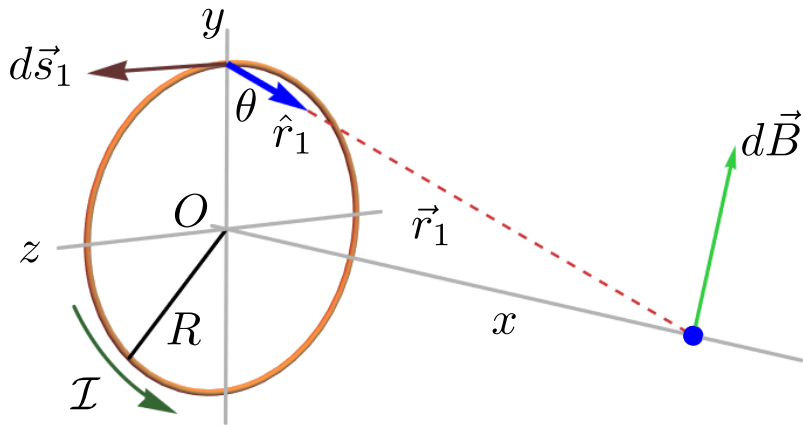


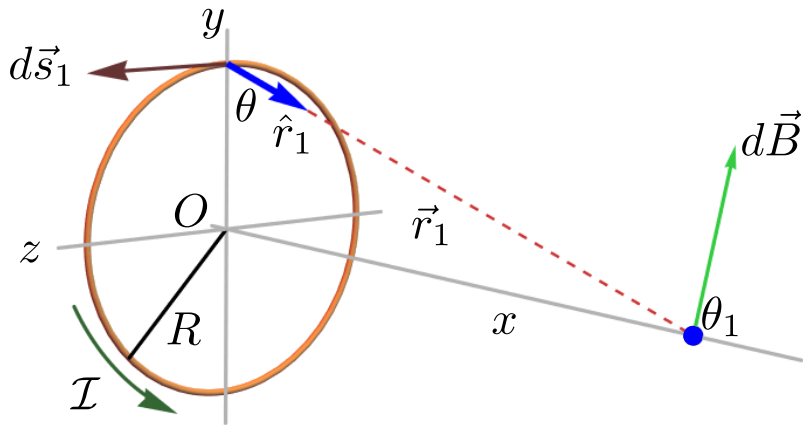
Demo is [here](#).

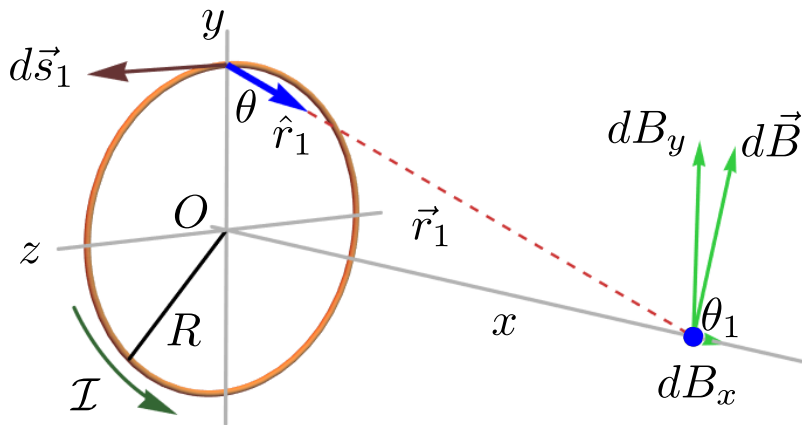


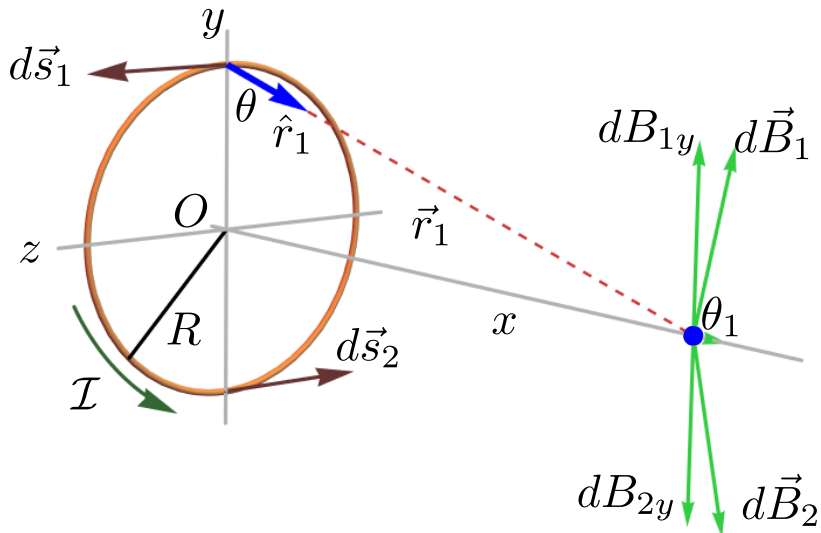


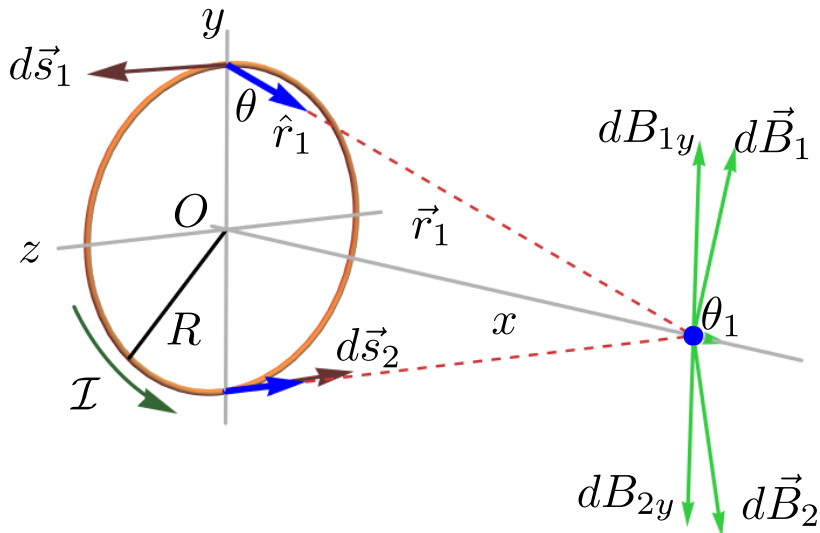




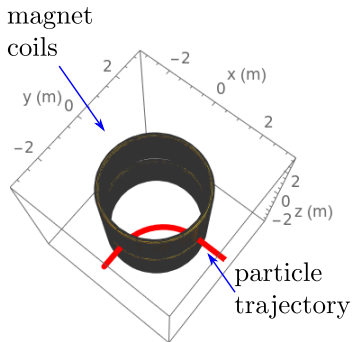
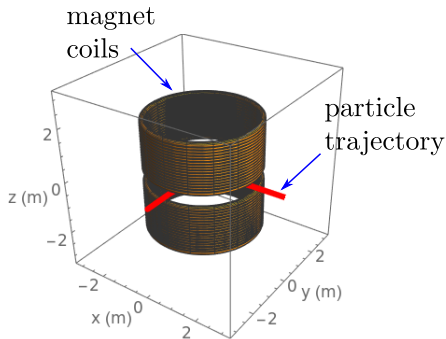






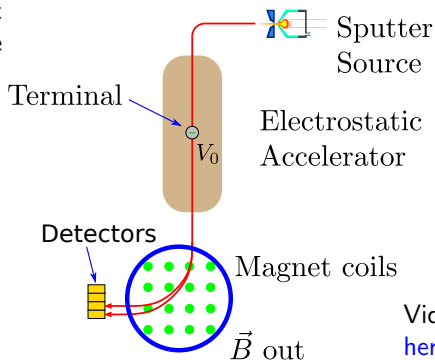


Usually many loops are used to reach the desired field as shown in the figure. What is the field for $R = 0.3 \text{ m}$, $I = 1116 \text{ A}$, an average distance from the center of each loop of $\langle x \rangle = 0.10 \text{ m}$, and the number of loops $N = 1116$? Assume the field inside the loops is the same as we calculated for the center and use $\langle x \rangle$ for the position along the center of the loop.

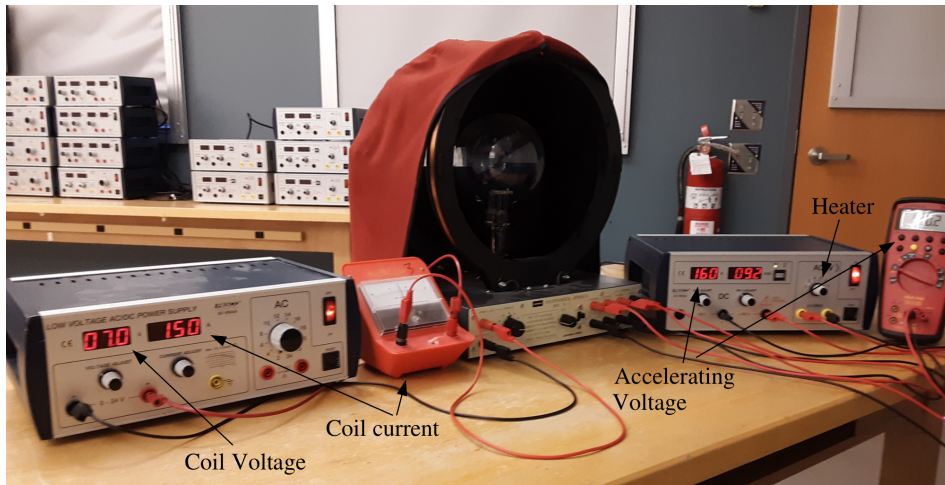


Taking the Ocean Temperature a Million Years Ago

Isotopes of oxygen, ^{16}O and ^{18}O , are accelerated twice across a potential difference $V_0 = 5.0 \times 10^6 \text{ V}$. Each atom carries an extra electron initially, but that electron and three others are stripped off the oxygen in the accelerator by a thin foil at the terminal. The oxygen has a charge $+3e$ and is moving horizontally when it enters a uniform magnetic field pointing straight up with $B = 3.0 \text{ T}$. The positively-charged atoms follow a circular path before striking detectors that measure their position. The variation in the trajectory of each atom leads to an uncertainty in their final position of about $\pm 0.007 \text{ m}$. Will the spectrometer be able to separate the isotopes?



Video is [here](#).



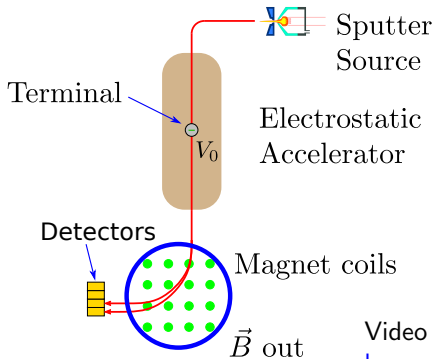
Demo is [here](#).

For Activities 5.b-5.d

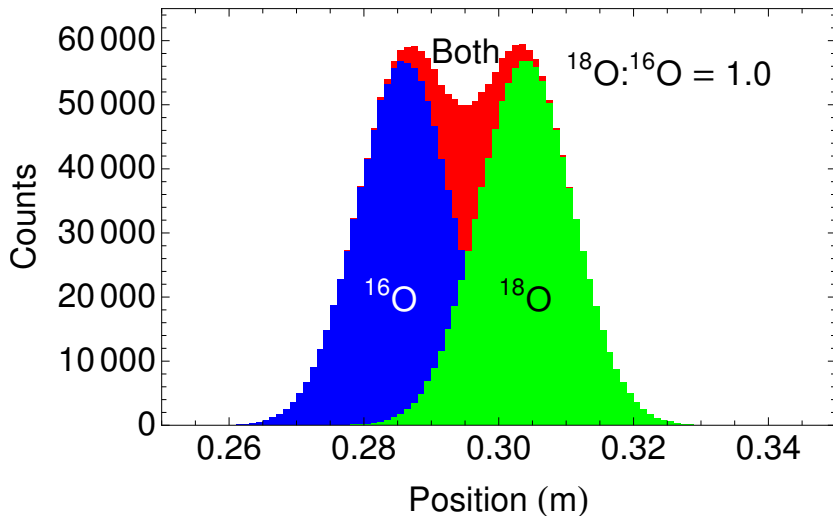
- 1 Get the electron mass m_e for each entry in the table.
- 2 Get the average and standard deviation for m_e .
- 3 State your final result for $m_e \pm \delta m_2$ in Activity 5.d. Does the accepted value fall within your range?

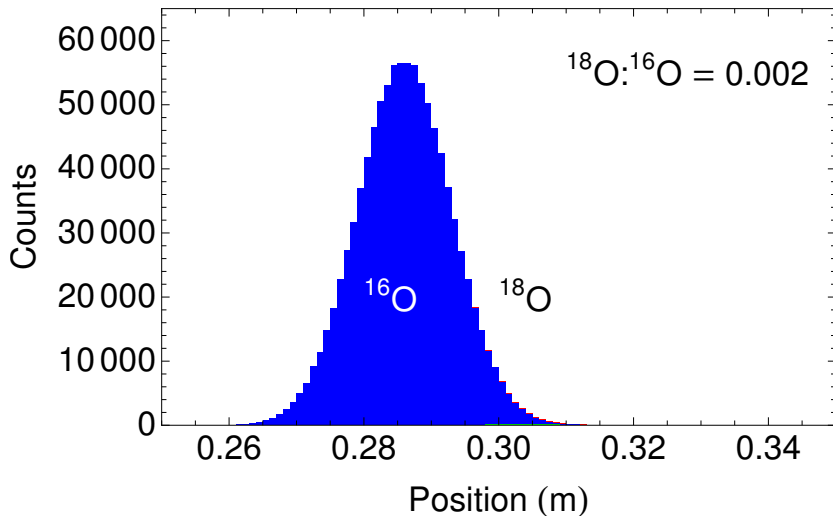
Taking the Ocean Temperature a Million Years Ago

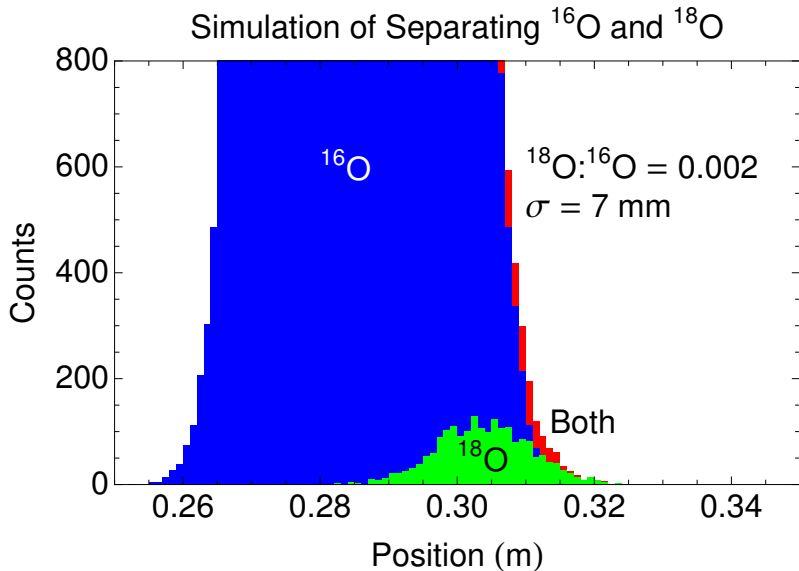
Oxygen isotopes, ^{16}O ($m_{16} = 2.67 \times 10^{-26} \text{ kg}$) and ^{18}O ($m_{18} = 3.0 \times 10^{-26} \text{ kg}$), are accelerated twice across a potential difference $V_0 = 5.0 \times 10^6 \text{ V}$. Each atom starts with an extra electron, but that electron and three others are stripped off the oxygen by a thin foil at the terminal. The oxygen has a charge $+3e$ and is moving horizontally when it enters a uniform magnetic field pointing straight up with $B = 3.0 \text{ T}$. The positively-charged atoms follow a circular path before striking detectors that measure their position. The variation in the trajectory of each atom leads to an uncertainty in their final position of about $\pm 0.007 \text{ m}$. Will the spectrometer be able to separate the isotopes?

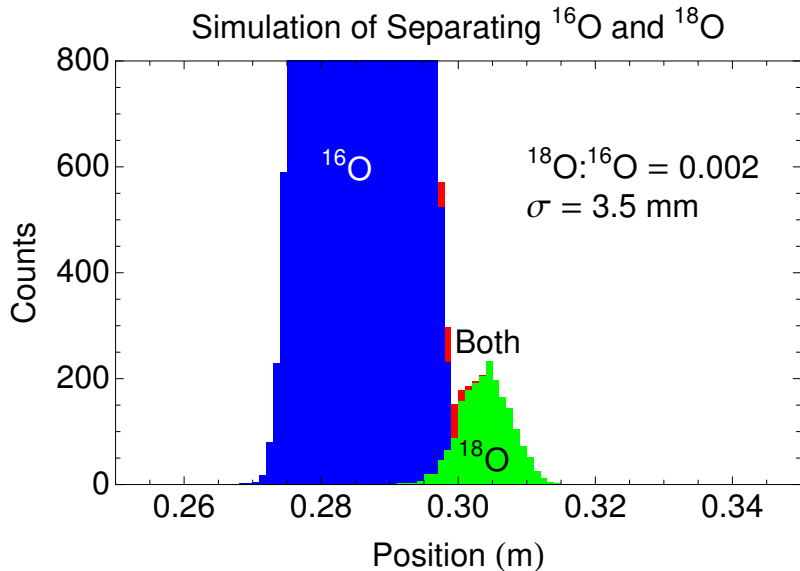


Video is [here](#).

Simulation of Separating ^{16}O and ^{18}O 

Simulation of Separating ^{16}O and ^{18}O 

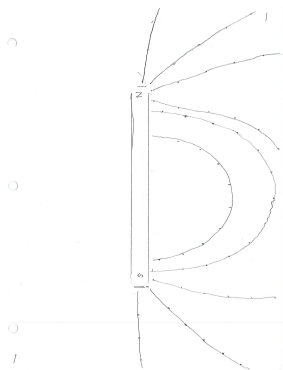


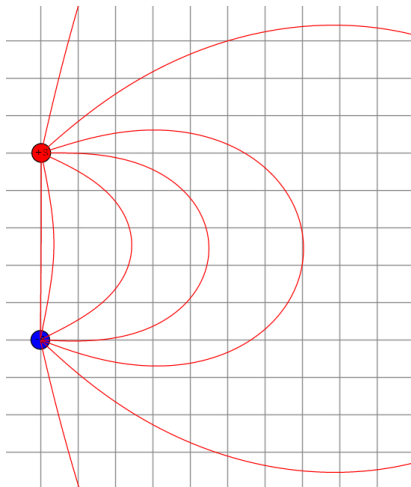
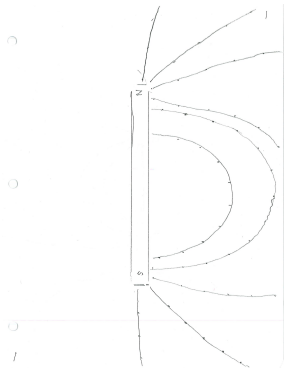




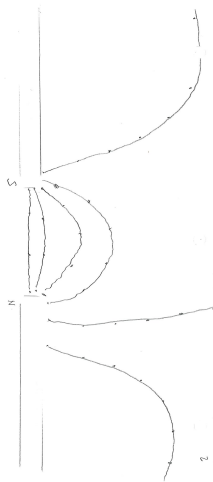
WE WERE GOING TO USE THE TIME MACHINE TO PREVENT THE ROBOT APOCALYPSE, BUT THE GUY WHO BUILT IT WAS AN ELECTRICAL ENGINEER.

Cueball tells Franklin that the charge left on a glass rod by rubbing it with silk should be the negative charge, not the positive charge, because the friction removes electrons from the rod. This would not have been intuitive to Franklin, because the electron had not as of yet been discovered. Yet by telling Franklin to reverse the positive and negative conventions, this would ultimately result in an alternate universe where electrons are assigned a positive charge. One can only speculate what other changes this reversal of convention would lead to, as small changes tend to cascade into huge ones. Would the positron have been instead named the negatron? And would this affect the success of the Transformers franchise?





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