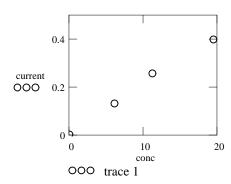
## Linear Calibration Example 2

conc := 
$$(0 \ 6.1 \ 11.2 \ 19.4)^{T}$$
 current :=  $(0.003 \ 0.134 \ 0.259 \ 0.398)^{T}$ 

Let's look at a plot of the calibration measurements.



Looks reasonably linear! Let's calculate the least-squares estimates of the slope and intercept of a best-fit line.

$$b_0 := intercept(conc, current)$$
  $b_0 = 0.0101$ 

$$b_1 := slope(conc, current)$$
  $b_1 = 0.0205$ 

We need to calculate the residuals and a few other items

$$fit := b_1 \cdot conc + b_0 \qquad res := current - fit$$

$$s_{res} := \sqrt{\frac{1}{2} \cdot \sum_{res}^{2}}$$
  $s_{res} = 0.0161$ 

$$S_{XX} := 3 \cdot Var(conc)$$
  $S_{XX} = 202.2875$   $xbar := mean(conc)$ 

Now we can calculate a confidence interval for the analyte concentration in the "unknown"

 $y_{11} := 0.175$ signal from the "unknown" sample, in microamps

$$x_u := \frac{y_u - b_0}{b_1}$$
  $x_u = 8.0307$  point estimate of analyte concentration, in ppb

$$se_{u} := \frac{s_{res}}{b_{1}} \cdot \sqrt{1 + \frac{1}{4} + \frac{\left(x_{u} - xbar\right)^{2}}{S_{xx}}}$$
 se  $u = 0.8797$  std error of point estimate, in ppb

$$t := qt(.975, 2)$$

$$t = 4.3027$$

$$t \cdot se_{11} = 3.7852$$

t := qt(.975, 2) t = 4.3027  $t \cdot se_{11} = 3.7852$  width of 95% confidence interval

The concentration of arsenic in the sample is  $8.0 \pm 3.8$  ppb (95% CL).

This confidence interval assumes homogeneous noise in the calibration measurements, and that the measurements are normally distributed.