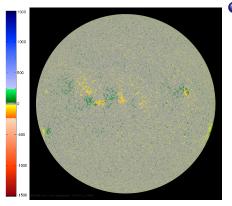
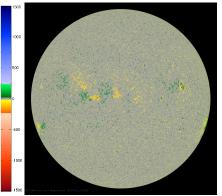


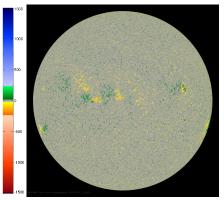
Int: It makes these happen.



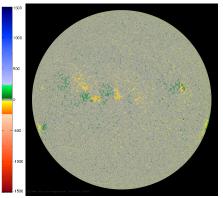
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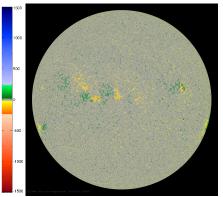
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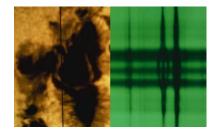


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- Why should you care? Ask him.

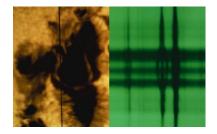
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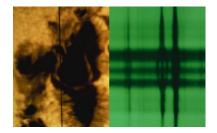
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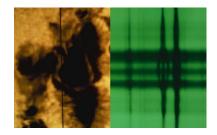
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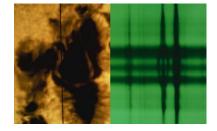
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The Zeeman Effect!

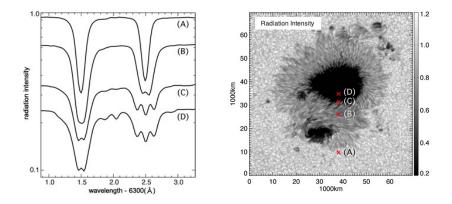




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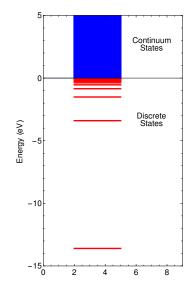
The Question

Left-hand plot below shows spectra of a sunspot. The labels from (A) to (D) correspond to the locations indicated in right-hand image. The single line (A) is split as the position shifts into the sunspot. The absorption lines at 6301.5 angstroms and 6302.5 angstroms are from iron. What is the size of the magnetic field that created the observed splitting?

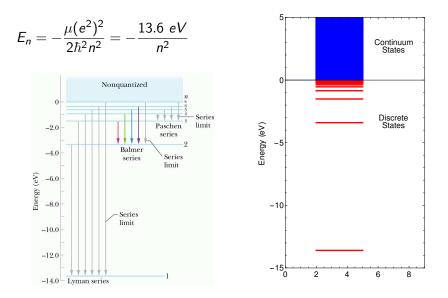


Hydrogen Eigenvalues (Energy Levels)

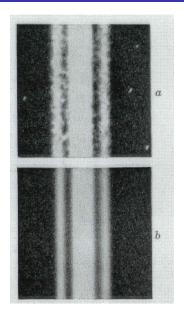
$$E_n = -\frac{\mu(e^2)^2}{2\hbar^2 n^2} = -\frac{13.6 \ eV}{n^2}$$



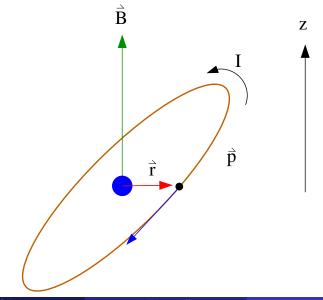
Hydrogen Eigenvalues (Energy Levels)

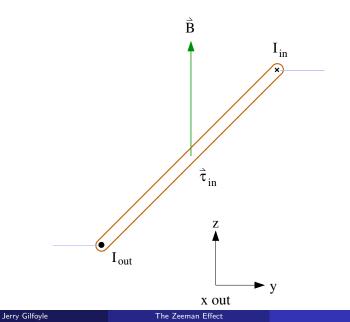


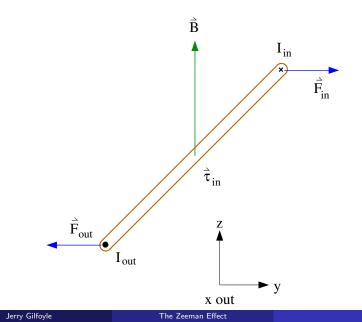
Zeeman Effect - H_{α} line in hydrogen



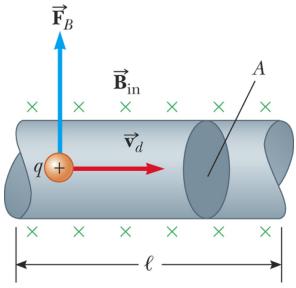
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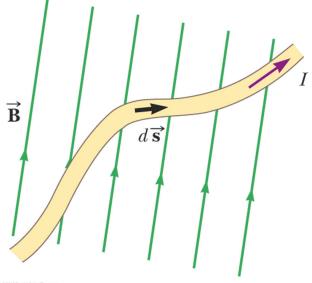




8 / 17

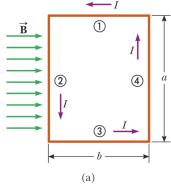


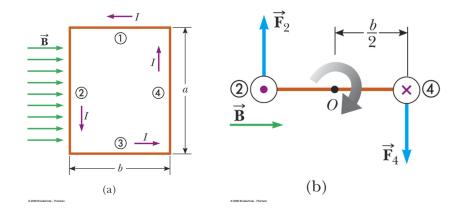
@ 2006 Brooks/Cole - Thomson

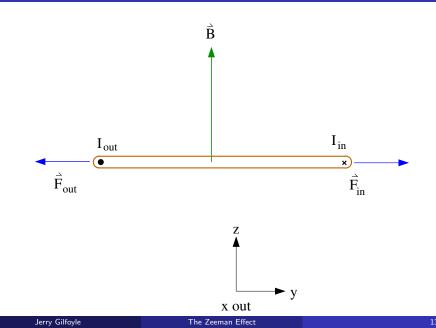


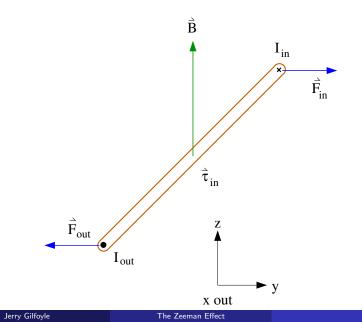
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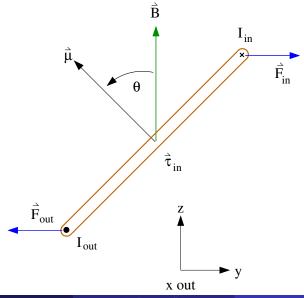
Jerry Gilfoyle







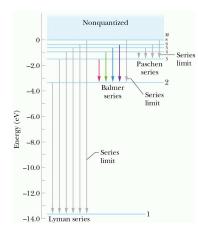


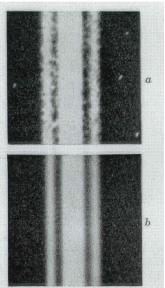


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The Zeeman Effect

Test the Model - Zeeman Effect and ${\rm H}_{\alpha}$ line in hydrogen





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