

# The Electromagnetic Properties of Materials

---

1. What are the types of electromagnetic matter?

# The Electromagnetic Properties of Materials

---

1. What are the types of electromagnetic matter?

- Conductors.

# The Electromagnetic Properties of Materials

---

1. What are the types of electromagnetic matter?

- Conductors.
- Insulators/Dielectrics.

# The Electromagnetic Properties of Materials

---

1. What are the types of electromagnetic matter?
  - Conductors.
  - Insulators/Dielectrics.
2. What happens when you put a neutral object in an electric field?

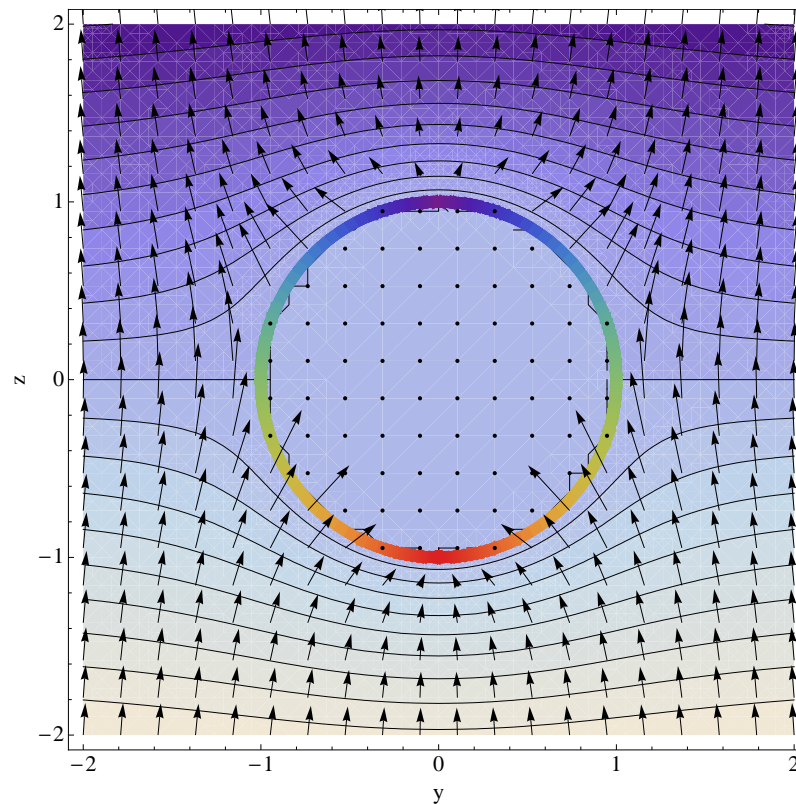
# The Electromagnetic Properties of Materials

---

1. What are the types of electromagnetic matter?

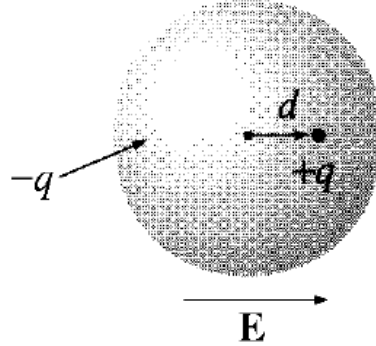
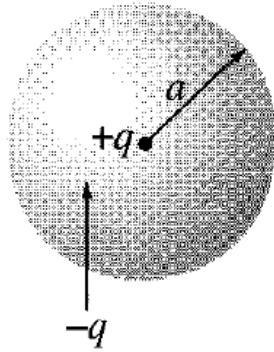
- Conductors.
- Insulators/Dielectrics.

2. What happens when you put a neutral object in an electric field?



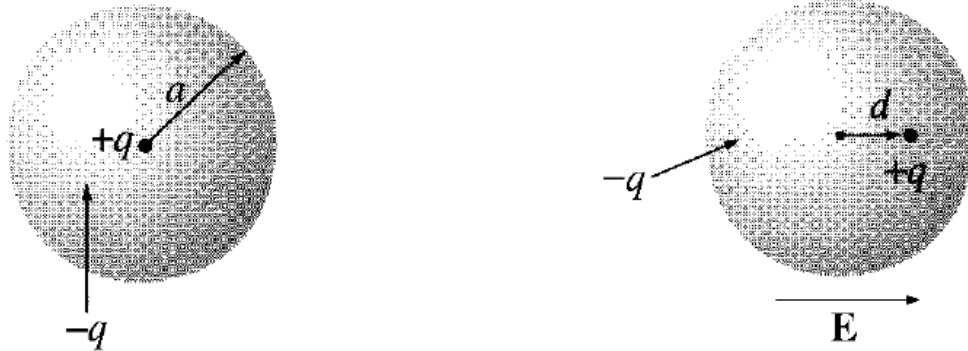
# Mechanisms for Polarizing Particles

- Induced Polarization: Shift electrons and nuclei.

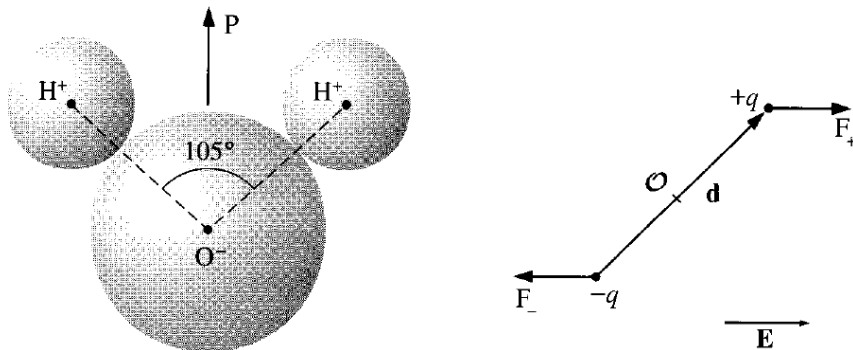


# Mechanisms for Polarizing Particles

- Induced Polarization: Shift electrons and nuclei.

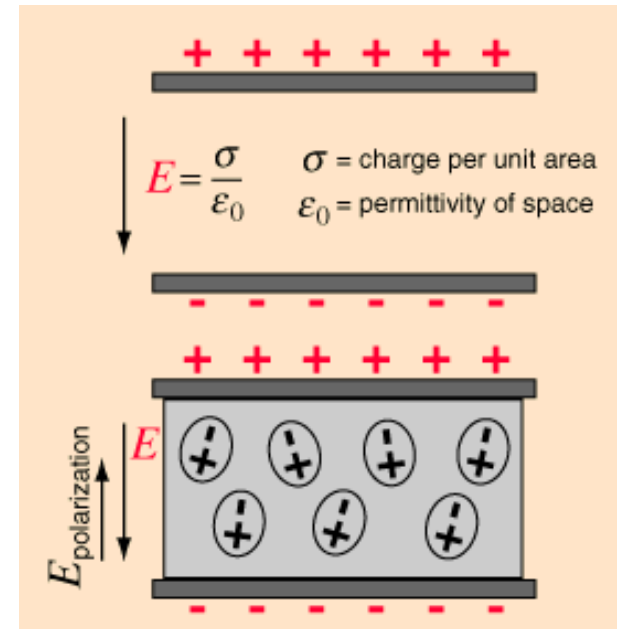


- Polar Objects: Some nuclei, molecules, *etc.* have a permanent dipole moment. They get rotated by the applied field.



# What is a dielectric?

- The dielectric constant of a material measures how the material responds to an applied external electric field.
- If the atoms in the material have a dipole moment they will tend to orient themselves in the applied field so the net field in the material is reduced.
- Internal electric field reduced from the vacuum value by the dielectric constant  $\epsilon_r$ .
- Dielectric are non-metallic substances (gas, liquid, or solid).
- Many practical applications like storing energy in capacitors, piezoelectric for making measurements, accumulating charge in an accelerator, *etc.*





# The Problem

---

In a linear dielectric, the polarization is proportional to the field. If the material is a gas of atoms, the induced dipole moment of each one is also proportional to the applied field. What is the relationship between the atomic polarizability  $\alpha$  and the dielectric constant  $\epsilon_r$ ? How well do your results agree with the data in the table below?

Gas	$\alpha/4\pi\epsilon_0$ ( $10^{-30} m^3$ )	$\epsilon_r^\dagger$
Hydrogen	0.667	1.00025
Helium	0.205	1.000065
Neon	0.396	1.00013
Argon	1.64	1.00052

# An Example

---

A thin, disk-shaped block of dielectric of thickness  $t$  and radius  $R$  has a uniform polarization  $\vec{P} = P\hat{z}$  as shown in the figure.

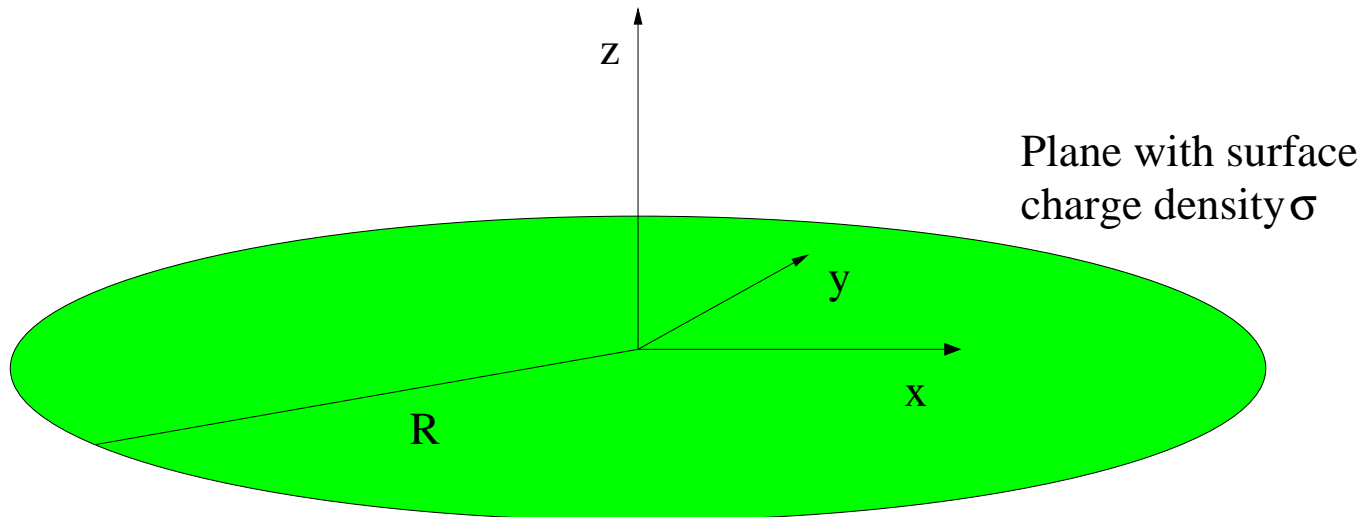
1. What are the bound charges  $\sigma_b$  and  $\rho_b$ ?
2. What is the field inside the dielectric along the axis of the disk?
3. What is the field outside the dielectric along the axis of the disk?



## Problem 2.6

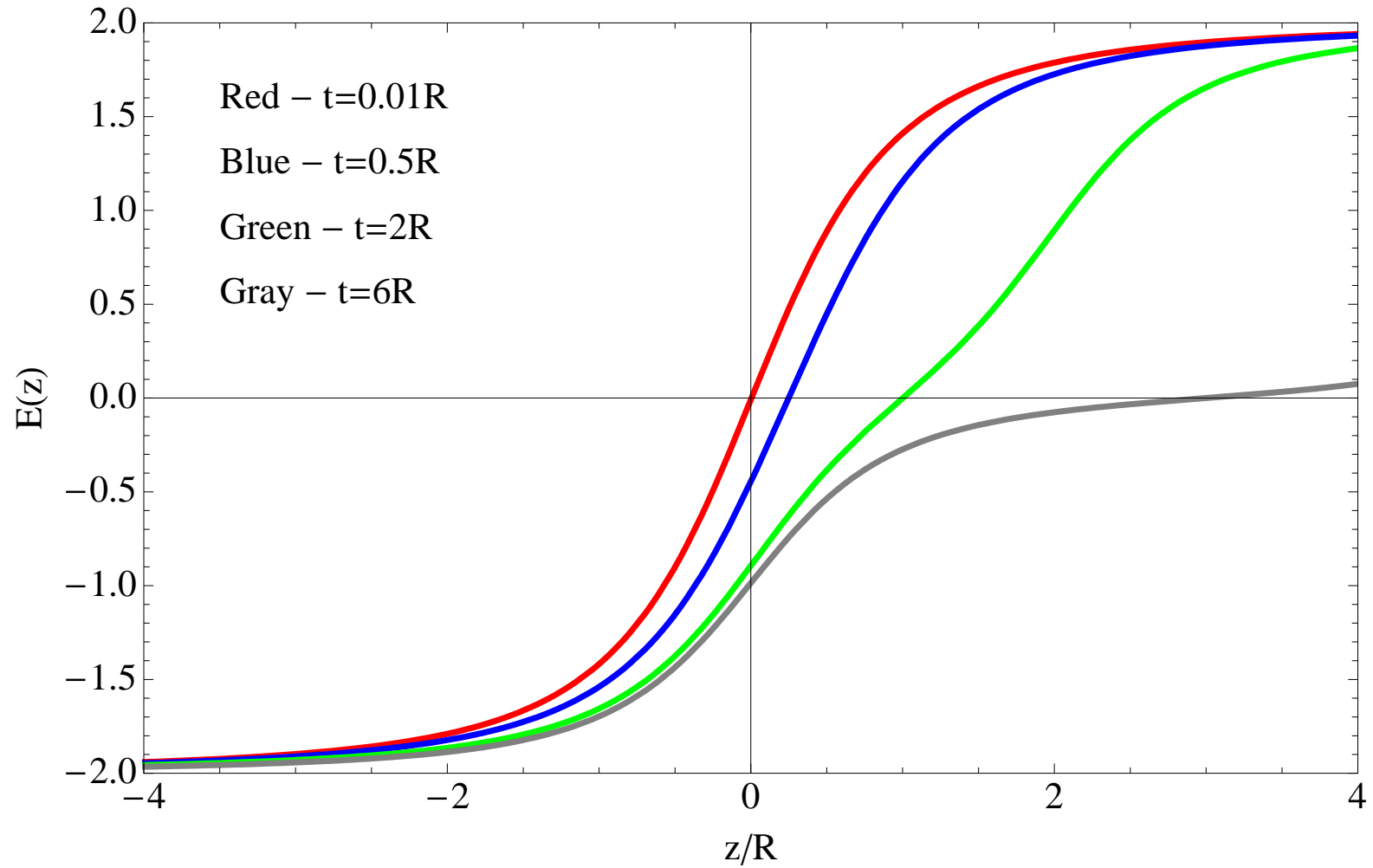
---

Find the electric field a distance  $z$  above the center of a flat circular disk of radius  $R$  (see figure below), which carries a uniform surface charge  $\sigma$ . What does your formula give in the limit  $R \rightarrow \infty$ ? Also check the case  $z \gg R$ .

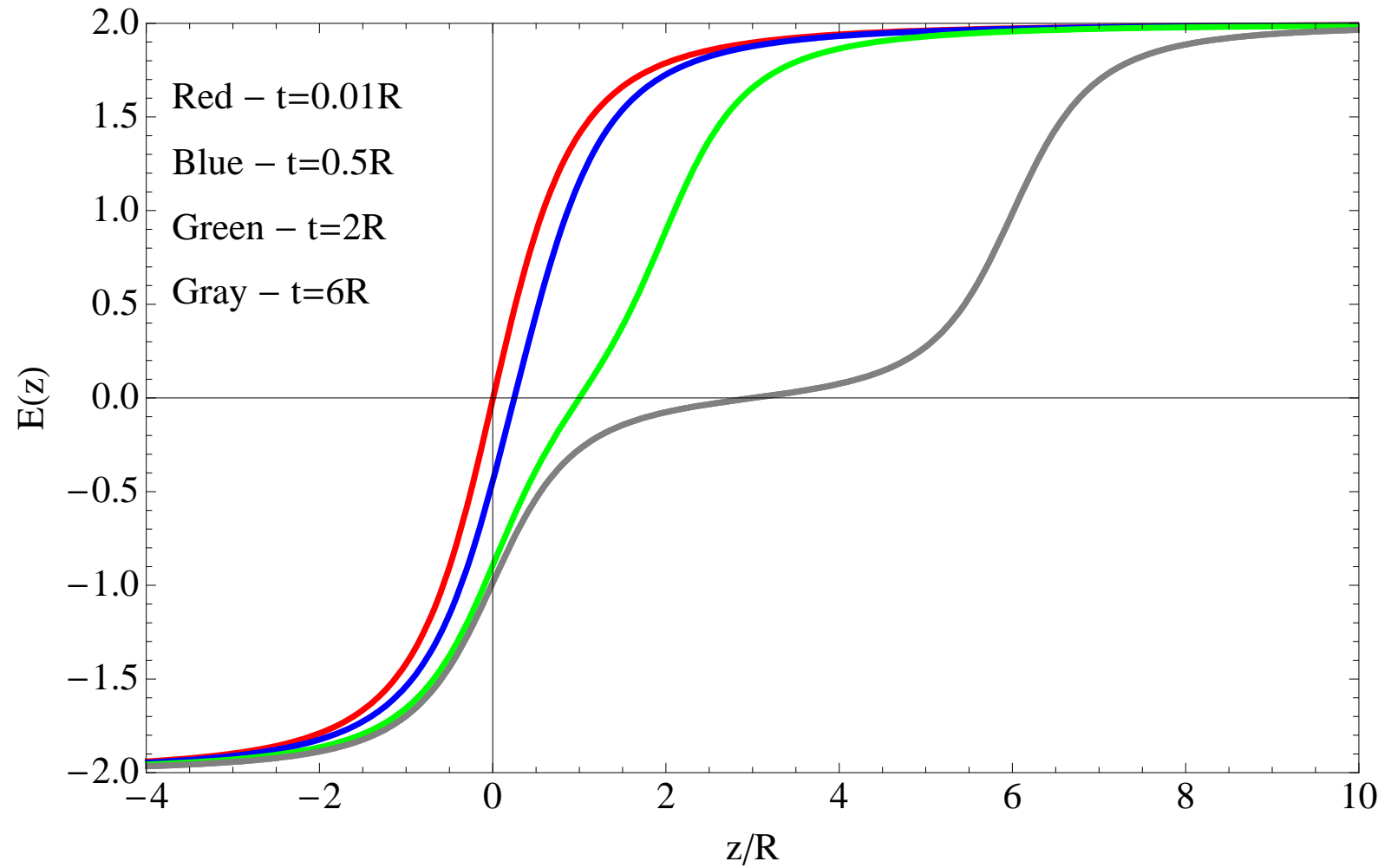


$$\vec{E} = \frac{1}{4\pi\epsilon_0} 2\pi\sigma z \left( \frac{1}{z} - \frac{1}{\sqrt{z^2 + R^2}} \right) \hat{z}$$

# Results for the Example



# Results for the Example



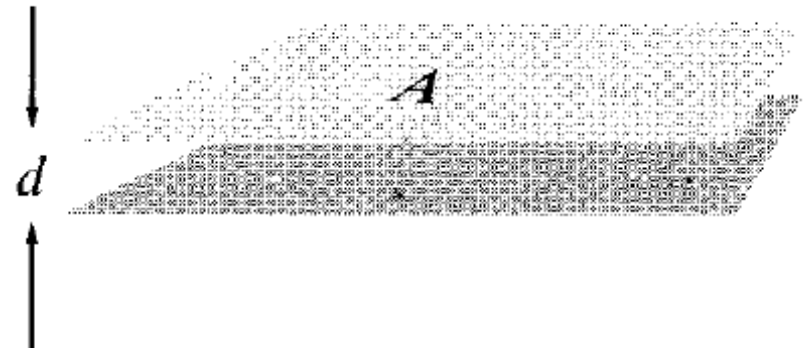
# Effect of a Dielectric On a Capacitor

A capacitor is a device for storing electromagnetic energy and has many uses in electronics, power generation and storage, *etc.* A parallel plate capacitor consists of two metal surfaces of area  $A$  held a distance  $d$  apart (see figure). The capacitance  $C$  is defined as

$$C = \frac{Q}{V}$$

where  $Q$  is the amount of charge on the positive plate,  $-Q$  is the amount on the negative plate, and  $V$  is the electric potential between the two plates. Assume  $d^2 \ll A$ .

1. What is the capacitance in terms of purely geometric parameters or constants?
2. Suppose you fill the space between the plates with an insulating material of dielectric constant  $\epsilon_r$ . What happens to the capacitance?
3. How is the energy stored in the capacitor effected?



# The Problem

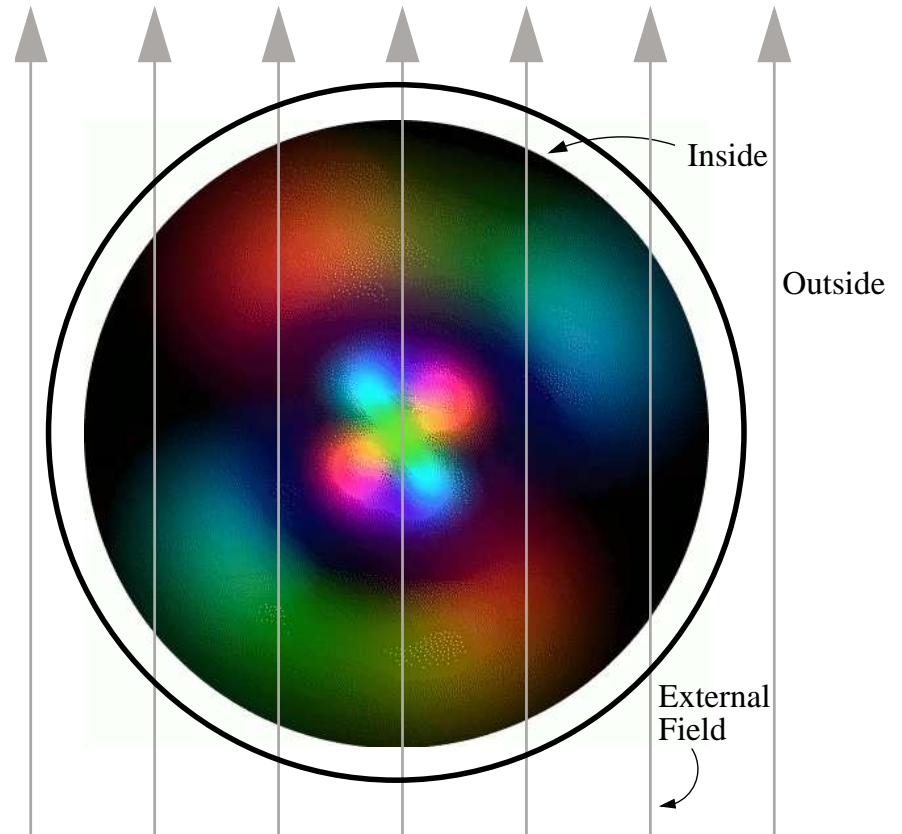
---

In a linear dielectric, the polarization is proportional to the field. If the material is a gas of atoms, the induced dipole moment of each one is also proportional to the applied field. What is the relationship between the atomic polarizability  $\alpha$  and the dielectric constant  $\epsilon_r$ ? How well do your results agree with the data in the table below?

Gas	$\alpha/4\pi\epsilon_0$ ( $10^{-30} m^3$ )	$\epsilon_r^\dagger$
Hydrogen	0.667	1.00025
Helium	0.205	1.000065
Neon	0.396	1.00013
Argon	1.64	1.00052

# A Picture of the Atomic Environment

1. The microscopic atomic environment is one of rapid change and sensitive dependence on distance, *i.e.* the electric fields change rapidly in time and space.
2. We are interested in understanding the bulk, macroscopic behavior of materials.
3. Go after the average properties of the material and the atomic environment.
4. Divide space into the region 'inside' where the atom is located and the 'outside' which is the average product of all the other atoms.

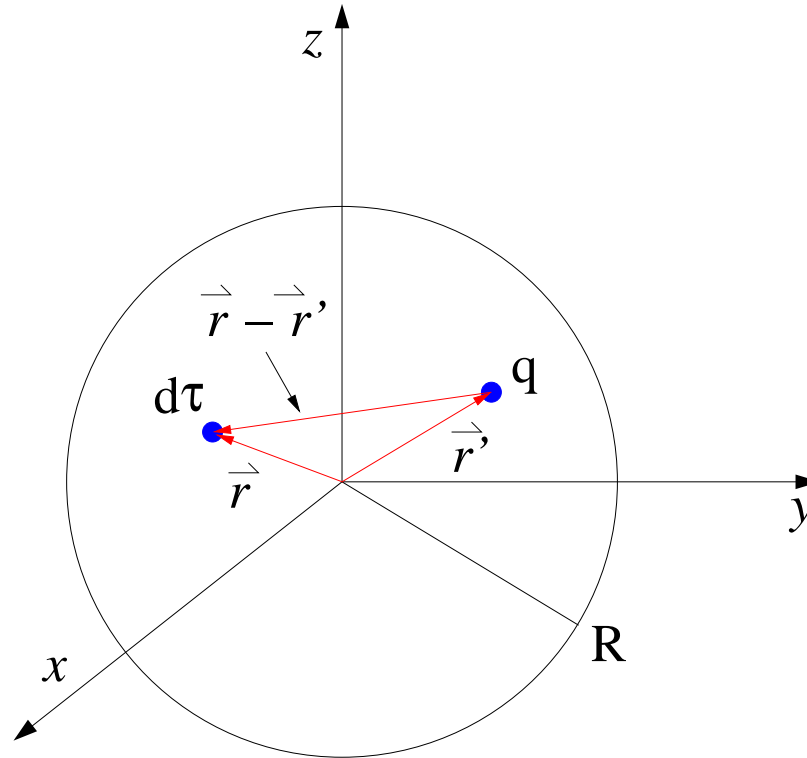




# Average Field of Our Spherical Atom

---

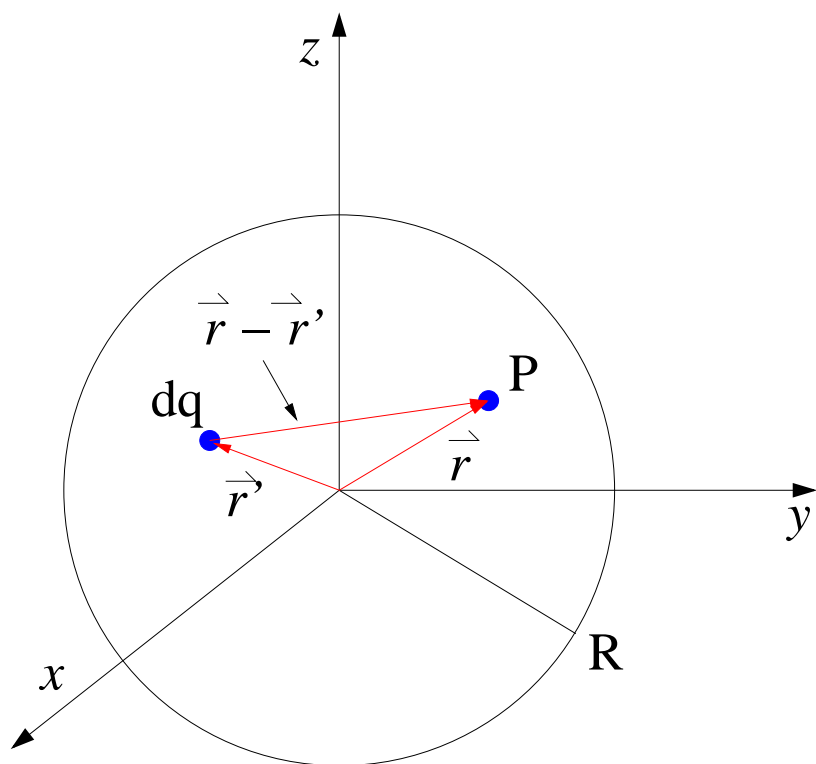
What is the average field inside a sphere of radius  $R$  due to all the charge within the sphere?



# The Field Inside a Uniform, Spherical, Charge Distribution

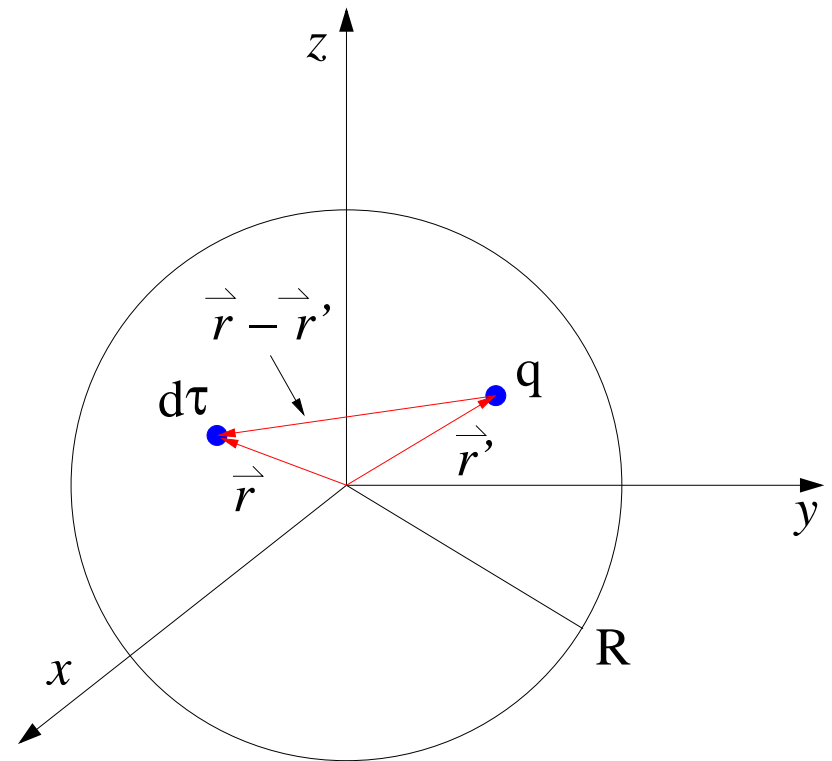
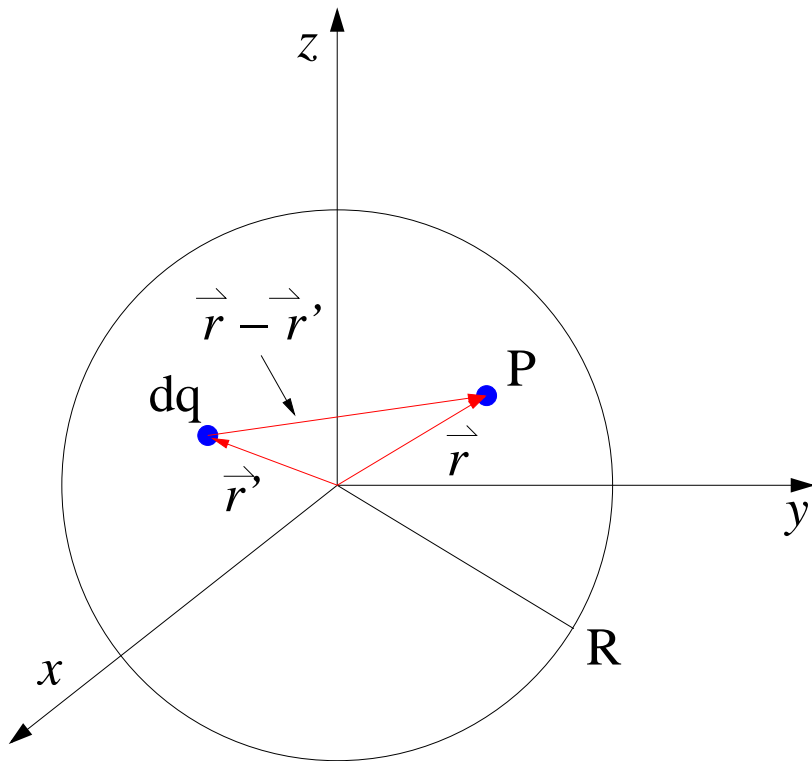
---

What is the electric field  $\vec{E}$  at a point  $\vec{r}$  inside a uniform sphere of charge with radius  $R$ , centered at the origin, and with  $r < R$ ?



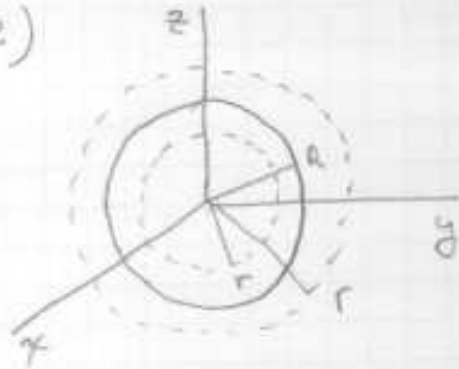
# The Field Inside a Uniform, Spherical, Charge Distribution

What is the electric field  $\vec{E}$  at a point  $\vec{r}$  inside a uniform sphere of charge with radius  $R$ , centered at the origin, and with  $r < R$ ?



# Problem 2-12

2-12)



for  $r > R$ , you get the same result as 2-11.

$$\vec{E}_{\text{out}} = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2} \hat{r} \quad \text{Same as 2.8}$$

for  $r < R$   $\nabla \cdot \vec{E}_{\text{in}} = \frac{\rho}{\epsilon_0}$

$$\int_V \nabla \cdot \vec{E}_{\text{in}} d\tau = \int_V \frac{\rho}{\epsilon_0} d\tau$$

$$\oint_S \vec{E}_{\text{in}} \cdot d\vec{A} = \frac{\rho}{\epsilon_0} \int_V d\tau$$

$$4\pi r^2 E_{\text{in}} = \frac{\rho}{\epsilon_0} \frac{4}{3}\pi r^3$$

$$E_{\text{in}} = \frac{\rho r}{3\epsilon_0}$$

$$\therefore \vec{E}_{\text{in}} = \frac{\rho r}{3\epsilon_0} \hat{r} \quad \text{Same as 2.8}$$

# The Electromagnetic Response of Atoms

---

For a gas of atoms the relationship between the atomic polarizability  $\alpha$  and the dielectric constant  $\epsilon_r$  is the following.

$$\epsilon_r = \frac{1 + \frac{2N\alpha}{3\epsilon_0}}{1 - \frac{N\alpha}{3\epsilon_0}}$$

Below is a comparison of the dielectric constants calculated from the measured  $\alpha$ 's.

Gas	$\alpha/4\pi\epsilon_0^*$	$\epsilon_r^\dagger$ (measured)	$\epsilon_r$ (calculated)
Hydrogen	0.667	1.00025	1.00023
Helium	0.205	1.000065	1.000071
Neon	0.396	1.00013	1.00014
Argon	1.64	1.00052	1.00056

\* Units of  $10^{-30} \text{ m}^3$ .

† For 1 atm, 20° C.

# The Electromagnetic Response of Atoms

For a gas of atoms the relationship between the atomic polarizability  $\alpha$  and the dielectric constant  $\epsilon_r$  is the following.

$$\epsilon_r = \frac{1 + \frac{2N\alpha}{3\epsilon_0}}{1 - \frac{N\alpha}{3\epsilon_0}}$$

Below is a comparison of the dielectric constants calculated from the measured  $\alpha$ 's.

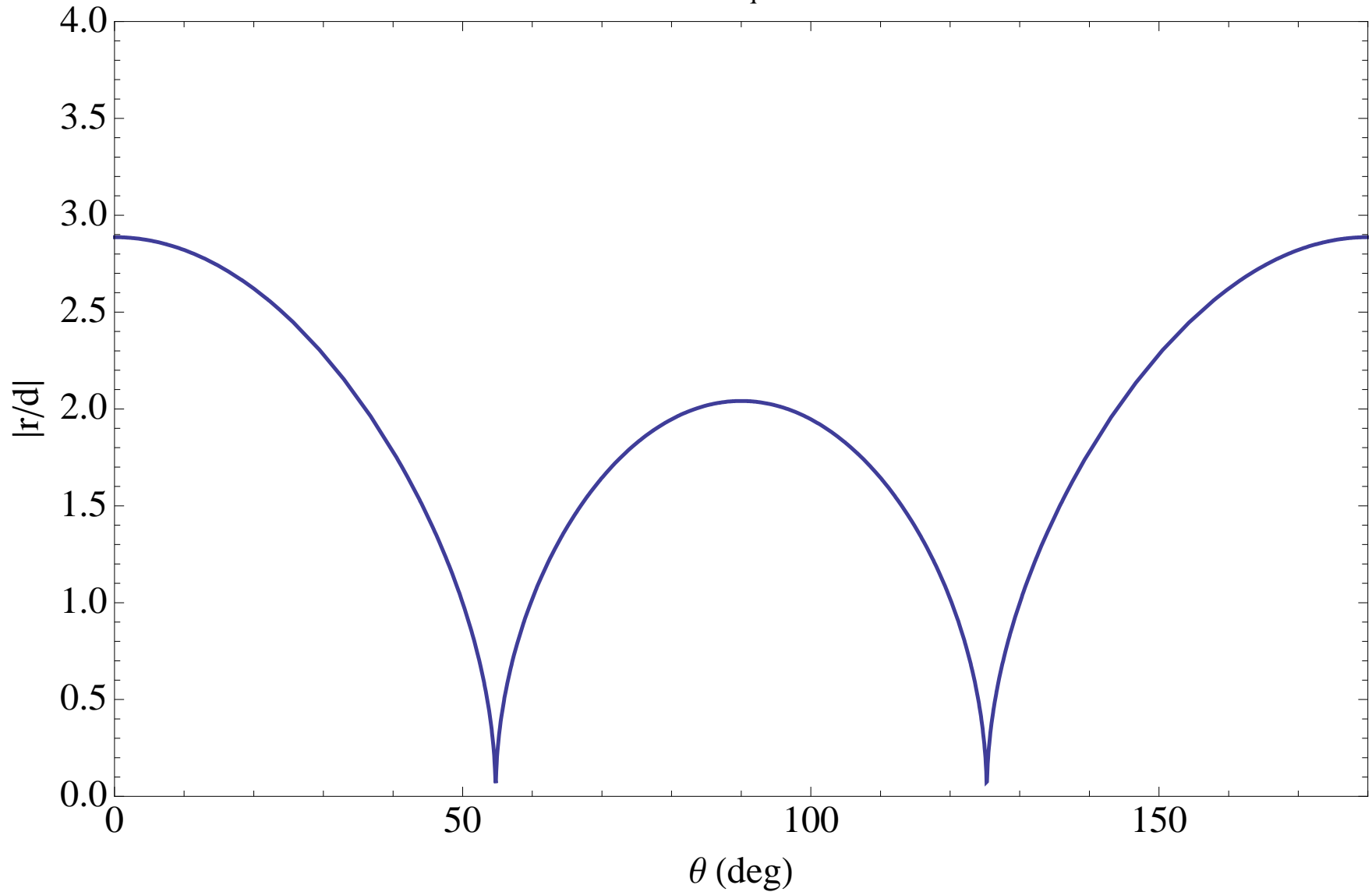
Gas	$\alpha/4\pi\epsilon_0^*$	$\epsilon_r^\dagger$ (measured)	$\epsilon_r$ (calculated)
Hydrogen	0.667	1.00025	1.00023
Helium	0.205	1.000065	1.000071
Neon	0.396	1.00013	1.00014
Argon	1.64	1.00052	1.00056

\* Units of  $10^{-30} \text{ m}^3$ .

† For 1 atm, 20° C.

# Electrostatics 4 Homework

Lower limit for  $V_{\text{quad}}/V_{\text{mono}} > 0.01$



# Electrostatics 4 Homework

Comparison of Line Charge Electric Potentials

